Stoichiometry Worksheet and Key

2 KClO₃ → 2 KCl + 3 O₂

How many moles of O₂ will be formed from 1.65 moles of KClO₃?

 $\begin{array}{c|cccc}
1.65 & \text{mol KClO}_3 & \underline{\qquad} & \text{mol O}_2 \\
& \underline{\qquad} & \text{mol KClO}_3 & \underline{\qquad} & \underline{\qquad} & \text{mol O}_2
\end{array}$

2. How many moles of KClO₃ are needed to make 3.50 moles of KCl?

3.50mol KCl = ___ mol KClO₃

3. How many moles of KCl will be formed from 2.73 moles of KClO₃?

4 Fe + 3 O₂ → 2 Fe₂O₃

4. How many moles of Fe₂O₃ are produced when 0.275 moles of Fe is reacted?

0.275 mol Fe = ___ mol Fe₂O₃

5. How many moles of Fe₂O₃ are produced when 31.0 moles of O₂ is reacted?

6. How many moles of O₂ are needed to react with 8.9 moles of Fe?

2 H₂O → 2 H₂ + O₂

7. How many moles of O₂ are produced when 1.26 moles of H₂O is reacted?

How many moles of H₂O are needed to produce 55.7 moles of H₂?

If enough H₂O is reacted to produce 3.40 moles of H₂, then how may moles of O₂ must have been made?
 (a bit challenging, but just think about it and you can probably figure it out)

Worksheet 5 1 Stoichiometry

John H. McNeill

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software A comprehensive and practical resource the text uses both Aspen Plus and Aspen Hysys simulation software The author describes the basic methodologies for computer aided design and offers a description of the basic steps of process simulation in Aspen Plus and Aspen Hysys The text reviews the design and simulation of individual simple unit operations that includes a mathematical model of each unit operation such as reactors separators and heat exchangers The author also explores the design of new plants and simulation of existing plants where conventional chemicals and material mixtures with measurable compositions are used In addition to aid in comprehension solutions to examples of real problems are included The final section covers plant design and simulation of processes using nonconventional components This important resource Includes information on the application of both the Aspen Plus and Aspen Hysys software that enables a comparison of the two software systems Combines the basic theoretical principles of chemical process and design with real world examples Covers both processes with conventional organic chemicals and processes with more complex materials such as solids oil blends polymers and electrolytes Presents examples that are solved using a new version of Aspen software ASPEN One 9 Written for students and academics in the field of process design Chemical Process Design and Simulation is a practical and accessible guide to the chemical process design and simulation using proven software SCORE High ICSE Chemistry Class 10 Maninder Kaur, 2025-08-27 The SCORE HIGH ICSE Chemistry for Class 10 is an all inclusive practice and revision companion crafted to enhance exam readiness It offers complete coverage of every chapter Periodic Table Bonding Acids Bases Salts Electrolysis Metallurgy Organic Chemistry and Numerical Problems strictly aligned with the latest ICSE curriculum The book contains a wide collection of solved short answer structured diagram based and numerical questions along with specimen paper patterns HOTS and competency based practice sets Step by step solutions common mistake alerts and examiner tips guide students towards accuracy and precision By targeting high scoring areas and practising with mock papers learners can build confidence and achieve excellent marks in ICSE Chemistry SELF-HELP TO ICSE CANDID CHEMISTRY 10 (SOLUTIONS OF EVERGREEN PUB.) 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provocative and relevant view of the topic and its importance to society and our daily lives Three themes are stressed throughout the text developing chemical thinking and a chemical vision learning problem solving methods and utilizing group work and discussion activities These themes involve and engage the students in their own learning processes they are

challenged to be active The presentation of topics has been altered to include a new chapter which introduces the students to scientific thinking and shows that chemistry involves interesting and relevant topics. The reorganization presents many core concepts in the first five chapters preparing students for later chapters In addition the author has added vignettes throughout the chapters referring to health technology the environment and society as well as to specific tools of direct use to students Characterizing Mechanisms of Simultaneous Biological Nutrient Removal During Wastewater Treatment P. F. Strom, 2004-01-01 Simultaneous biological nutrient removal SBNR is the removal of nitrogen and or phosphorus in excess of that required for biomass synthesis in biological wastewater treatment systems where there are no defined anaerobic and or anoxic zones The hypothesis is that one or more of three mechanisms is responsible within individual systems variations in the bioreactor macroenvironment created by the mixing pattern gradients within the floc microenvironment and or novel microorganism activity Understanding of the mechanisms of SBNR can be expected to lead to improved efficiency and reliability in its application Preliminary work documented SBNR in 7 full scale OrbalTM closed loop bioreactors A batch assay demonstrated that novel microorganism activity was of little importance in SBNR at the three plants tested While the floc microenvironment likely plays an important role in nitrogen removal in such plants it cannot explain phosphorus removal A computational fluid dynamics CFD model was developed to elucidate the role of the bioreactor macroenvironment in SBNR This is the first reported application of CFD to activated sludge biological wastewater treatment Although the software and computational requirements limited model complexity it still simulated the creation of dissolved oxygen gradients within the system demonstrating that the anaerobic zones required for SBNR could occur

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