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39/40 GANIU ODUBANWO STREET POTA BADAGRY

THIRD TERM EXAMINATION

SUBJECT: CHEMISTRY

CLASS: S. S. S ONE

DURATION: 2HR

1. Acid turns blue litmus paper \_\_\_\_\_?  
a. yellow    b. colourless    c. green    d. red
2. Acids are characterized by    a. Bitter taste    b. soapy taste  
c. sour taste    d. all listed
3. An element existing in two or more physical states and maintaining the same chemical characteristics is called \_\_\_\_\_ a. polymerization    b. allotropes  
c. isotopes    d. crystal
4. Which allotrope of carbon is used in lead pencil?  
a. graphite    b. coke    c. coal    d. diamond
5. Carbon dioxide is a gas that burns with \_\_\_\_\_ flame  
a. yellow    b. green    c. blue    d. colourless
6. Hydrocarbons are of two classes namely: a. graphite and diamond  
b. coke and coal    c. aliphatic and gas    d. aromatic and aliphatic
7. An example of aromatic hydrocarbons is    a. Ethane    b. methane  
c. cyclopropane    d. benzene
8. Which is not a raw material for chemical industry    a. petroleum  
b. coal    c. alum    d. salt
9. Which is not a use of hydrogen trioxocarbonate    a. As baking powder  
b. As health salts    b. As fire extinguishers    d. To manufacture hydroxide
10. Which is not a property of carbon(II)oxide    a. It is colourless  
b. It is slightly lighter than air    c. It is insoluble in water    d. It is soluble
11. Which is a type of amorphous    a. carbon black charcoal    b. plant charcoal  
c. red charcoal    d. graded charcoal
12. Which of these is not a type of coal?    a. peat    b. lignite bituminous    c. coal
13. Coke is a non-volatile residue, this statement is    a. true    b. false    c. negative
14. Water gas is a mixture that contains Co and ..... gas?    a. hydrogen  
b. helium    c. lithium    d. berlium
15. The number of replaceable hydrogen atom in one molecule of an acid indicates its  
a. Basicity    b. Alkalinity    c. Acidity    d. Reactivity
16. Which of the following is /are characteristics properties of acids? They i. form hydroxonium ions in water ii. liberate carbon (iv)oxide from trioxocarbonate(iv) salts. iii. Turn blue litmus paper red. iv. have a soapy feel  
a. i, iv only    b. i and ii    c. iii and iv    d. i, ii, and iii
17. The following acid are monobasic except    a. Trioxonitrate(v)acid    b. Hydrochloric acid  
c. Ethanoic acid    d. Tetraoxosulphate (v)acid
18. Which of the following is an acid salt?    a.  $(\text{NH}_4)_2\text{CO}_3$     b.  $\text{CH}_3\text{COONa}$     c.  $\text{KHSO}_4$     d.  $\text{MgSO}_4$
19. Alums are classified as    a. basic salts    b. acid salts    c. anhydrous salts    d. double salts
20. Which of the following conclusion about solution of pH4 is correct  
a. it contains more  $\text{OH}^-$  than  $\text{H}_3\text{O}^+$     b. its POH value will be 10    c. it is more acidic than a solution of pH2

# Ss1 Chemistry Practical 2015 3rd Term

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