



Time: 1.30 Hours

Standard 12

CHEMISTRY

Part - I

Marks: 50

I. Choose the correct answer

10×1=10

- A complex in which the oxidation number of the metal is zero is
 - $K_4[Fe(CN)_6]$
 - $[Fe(CN)_3(NH_3)_3]$
 - $[Fe(CO)_5]$
 - both (b) and (c)
- Which of the following is paramagnetic in nature
 - $[Zn(NH_3)_4]^{2+}$
 - $[Co(NH_3)_6]^{3+}$
 - $[Ni(H_2O)_6]^{2+}$
 - $[Ni(CN)_4]^{2-}$
- A magnetic moment of 4.89 BM will be shown by one among the following
 - $TiCl_4$
 - $[CoF_6]^{3-}$
 - $[Ni(CN)_4]^{2-}$
 - $[Cu(NH_3)_4]^{2+}$
- While charging lead storage battery
 - $PbSO_4$ on cathode is reduced to Pb
 - $PbSO_4$ on anode is oxidised to PbO_2
 - $PbSO_4$ on anode is reduced to Pb
 - $PbSO_4$ on cathode is oxidised to Pb
- Among the following cells

I) Leclanche cell	II) Nickel - cadmium cell
III) Lead storage battery	IV) Mercury cell primary cells are
a) I and IV	c) III and IV
b) I and III	d) II and III
- Which of the following amines does not undergo acetylation?
 - t-butylamine
 - ethylamine
 - diethylamine
 - triethylamine
- Which one of the following will not undergo Hofmann bromide reaction
 - $CH_3CONHCH_3$
 - $CH_3CH_2CONH_2$
 - CH_3CONH_2
 - $C_6H_5CONH_2$
- Secondary nitro alkanes react with nitrous acid to form
 - red solution
 - blue solution
 - green solution
 - yellow solution
- The product formed by the reaction an aldehyde with a primary amine
 - carboxylic acid
 - aromatic acid
 - Schiff's base
 - ketone
- How many faradays of electricity are required for the following reaction to occur $MnO_4^- \rightarrow Mn^{2+}$
 - 5 F
 - 3 F
 - 1 F
 - 7 F



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M Woodhall

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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