

রসায়ন
সৃজনশীল প্রশ্ন

বিষয় কোড : ১ ৩ ৭

দ্রষ্টব্য : কোন পদার্থের সংখ্যা প্রদেয় পূর্ণমান প্রাপক। প্রদত্ত উদ্দীপকগুলো মনোযোগ দিয়ে পড় করো এবং সংশ্লিষ্ট প্রশ্নগুলোর সমাধান উত্তর দাও। যে কোনো ভুলটি এসএসসির উত্তর দাও।

১. (i) $\text{NaOH} + \text{HCl} = \text{NaCl} + \text{H}_2\text{O}$
 (ii) $\text{NaCl} + \text{AgNO}_3 = \text{AgCl} \downarrow + \text{NaNO}_3$
 (iii) $\text{SnCl}_2 + 2\text{FeCl}_3 = \text{SnCl}_4 + 2\text{FeCl}_2$

- ক. লিমিটিং বিক্রিয়ক কাকে বলে? ১
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২. একটি যৌগ কার্বন, হাইড্রোজেন এবং অক্সিজেন দ্বারা গঠিত যৌগটিতে $\text{C} = 40\%$, $\text{H} = 6.67\%$ যৌগটির আণবিক ভর 180।

- ক. Stoichiometry কী? ১
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- ক. পোলার যৌগ কী? ১
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৪. ▶

পর্যায় → 3
 4
 5

Na	X	Y	Si	P	S	Z
K						Cl
Rb						Br
						I

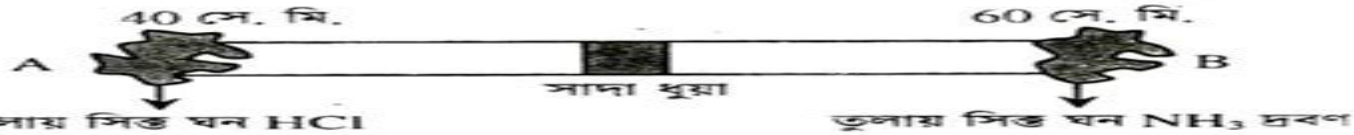
- (X, Y, Z প্রচলিত কোন মৌলের প্রতীক নয়)

- ক. ম্যান্ডেলিফের সংশোধিত পর্যায় সূত্রটি লিখ। ১
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৫. (i) C_2H_4 (ii) $\text{CH}_3 - \text{CH}_2\text{OH}$ (iii) $\text{CH}_3 - \text{COOH}$

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