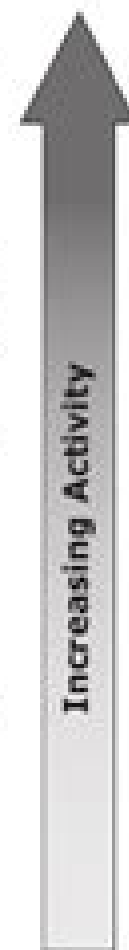


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POLYATOMIC IONS		SOLUBILITY OF COMMON IONIC COMPOUNDS IN WATER		ACTIVITY SERIES
Acetate	$\text{C}_2\text{H}_3\text{O}_2^-$, CH_3COO^-	<u>Soluble</u> <u>compounds contain</u> $\text{C}_2\text{H}_3\text{O}_2^-$, CH_3COO^-	<u>Common exceptions</u> None	<u>Metal</u> Lithium
Ammonium	NH_4^+	NH_4^+	None	Potassium
Carbonate	CO_3^{2-}	NO_3^-	None	Barium
Chlorate	ClO_3^-	CN^-	None	Calcium
Chlorite	ClO_2^-	ClO^-	None	Sodium
Chromate	CrO_4^{2-}	CO_3^{2-}	None	Magnesium
Cyanide	CN^-	ClO_2^-	None	Aluminum
Dichromate	$\text{Cr}_2\text{O}_7^{2-}$	ClO_4^-	None	Manganese
Hydrogen carbonate	HCO_3^-	Br^-	Compounds of Ag^+ , Pb^{2+} , and Hg_2^{2+}	Zinc
Hydroxide	OH^-	Cl^-	Compounds of Ag^+ , Pb^{2+} , and Hg_2^{2+}	Chromium
Hypochlorite	ClO^-	I^-	Compounds of Ag^+ , Pb^{2+} , and Hg_2^{2+}	Iron
Nitrate	NO_3^-	SO_4^{2-}	Compounds of Sr^{2+} , Ba^{2+} , Pb^{2+} , and Hg_2^{2+}	Cobalt
Nitrite	NO_2^-	<u>Insoluble</u> <u>compounds contain</u> CO_3^{2-}	<u>Common exceptions</u> Compounds of NH_4^+ and the alkali metal cations	Nickel
Perchlorate	ClO_4^-	PO_4^{3-}	Compounds of NH_4^+ and the alkali metal cations	Tin
Permanganate	MnO_4^-	CrO_4^{2-}	Compounds of NH_4^+ and the alkali metal cations	Lead
Phosphate	PO_4^{3-}	$\text{Cr}_2\text{O}_7^{2-}$	Compounds of NH_4^+ and the alkali metal cations	(Hydrogen)
Sulfate	SO_4^{2-}	OH^-	Compounds of NH_4^+ , the alkali metal cations, Ca^{2+} , Sr^{2+} , and Ba^{2+}	Copper
Sulfite	SO_3^{2-}	S^{2-}	Compounds of NH_4^+ , the alkali metal cations, Ca^{2+} , Sr^{2+} , and Ba^{2+}	Mercury
				Silver
				Platinum
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