ENHANCING TEST ITEMS - CHEMISTRY (for Physical Sciences majors)

- 1. Which of the following statements best describes the atomic radius of potassium atom and the atomic radius of a calcium atom?
 - A. The radius of the potassium atom is smaller because of its smaller nuclear charge.
 - B. The radius of the potassium atom is smaller because of its larger nuclear size.
 - The radius of the potassium atom is larger because its nuclear charge is smaller.
 - D. The radius of the potassium atom is larger because its nuclear charge is also larger.
- 2. Which type of chemical reaction occurs when an iron nail rusts?
 - A. condensation
 - B. neutralization

- C. oxidation-reduction
- D. ionization-dissociation
- 3. Which electron dot symbol represents the atom in period 4 with the highest first ionization en engy?











- 4. Which phase change is exothermic?
 - A. H₂O₁₀ --- H₂O₁₀
 - B. H₂O_m -+ H₂O_m

- Given the diagram, which statement BEST describes the key?
 - It is negative and acts as the cathode.
 - B. It is positive and acts as the cathode.
 - C. It is negative and acts as the anode.
 - D. It is positive and act as the anode
- 6. Which technique is safest for diluting a concentrated acid with water?
 - A. Add the acid to the water quickly.
 - B. Add the water to the acid quickly.
 - Add the acid to the water slowly while stirring constantly.
 - D. Add the water to the acid slowly while stirring constantly.
- 7. Which elements combine to form an ionic bond?
 - sodium and potassium
- C. carbon and oxygen

8. sodium and oxygen

- D. carbon and sulfur
- 8. As the elements in group 16 are considered in order from top to bottom, the atomic radius of each successive element increases. This increase is primarily due to an increase in
 - A. electronegativity

C. the number of protons occupying the nucleus

B. ionization energy

- D. the number of occupied principal energy levels
- 9. An air pollution situation in which reactions occur on the surfaces of solid particles and produce acidic irritation of lung tissues is called .

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Phillip John Buddle, J. S. Mackay

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