

14

THE BEHAVIOR OF GASES

Chapter Quiz

Indicate whether the statement is true or false.

1. Boyle's law states that, at constant temperature, the volume of a fixed amount of gas is inversely proportional to the pressure.
2. According to Charles's law, the volume of a fixed amount of gas is directly proportional to the absolute temperature.
3. Gay-Lussac's law states that, at constant volume, the pressure of a fixed amount of gas is directly proportional to the absolute temperature.
4. The average molecular mass of gases is proportional to the density of the gas.
5. The average molecular mass of gases is proportional to the square of the root-mean-square speed of the gas molecules.
6. The average molecular mass of gases is proportional to the square of the root-mean-square speed of the gas molecules.

Indicate the correct answer for the question.

7. A gas is compressed from 1.00 atm to 2.00 atm at a constant temperature. What is the new volume?

Answers:

1. True. Boyle's law states that, at constant temperature, the volume of a fixed amount of gas is inversely proportional to the pressure.
2. True. According to Charles's law, the volume of a fixed amount of gas is directly proportional to the absolute temperature.
3. True. Gay-Lussac's law states that, at constant volume, the pressure of a fixed amount of gas is directly proportional to the absolute temperature.
4. True. The average molecular mass of gases is proportional to the density of the gas.
5. True. The average molecular mass of gases is proportional to the square of the root-mean-square speed of the gas molecules.
6. True. The average molecular mass of gases is proportional to the square of the root-mean-square speed of the gas molecules.

Answers:

7. 0.500 L. A gas is compressed from 1.00 atm to 2.00 atm at a constant temperature. What is the new volume? The initial volume is 1.00 L. The final volume is 0.500 L.

1. Boyle's law states that, at constant temperature, the volume of a fixed amount of gas is inversely proportional to the pressure.	True
2. According to Charles's law, the volume of a fixed amount of gas is directly proportional to the absolute temperature.	True
3. Gay-Lussac's law states that, at constant volume, the pressure of a fixed amount of gas is directly proportional to the absolute temperature.	True
4. The average molecular mass of gases is proportional to the density of the gas.	True
5. The average molecular mass of gases is proportional to the square of the root-mean-square speed of the gas molecules.	True
6. The average molecular mass of gases is proportional to the square of the root-mean-square speed of the gas molecules.	True

True

True

True

The Behavior Of Gases Chapter Quiz

Minjie Lin



The Behavior Of Gases Chapter Quiz:

The Top Books of the Year The Behavior Of Gases Chapter Quiz The year 2023 has witnessed a remarkable surge in literary brilliance, with numerous engrossing novels captivating the hearts of readers worldwide. Lets delve into the realm of popular books, exploring the fascinating narratives that have charmed audiences this year. The Behavior Of Gases Chapter Quiz : Colleen Hoover's "It Ends with Us" This heartfelt tale of love, loss, and resilience has captivated readers with its raw and emotional exploration of domestic abuse. Hoover masterfully weaves a story of hope and healing, reminding us that even in the darkest of times, the human spirit can prevail. The Behavior Of Gases Chapter Quiz : Taylor Jenkins Reid's "The Seven Husbands of Evelyn Hugo" This captivating historical fiction novel unravels the life of Evelyn Hugo, a Hollywood icon who defies expectations and societal norms to pursue her dreams. Reid's absorbing storytelling and compelling characters transport readers to a bygone era, immersing them in a world of glamour, ambition, and self-discovery. The Behavior Of Gases Chapter Quiz : Delia Owens' "Where the Crawdads Sing" This mesmerizing coming-of-age story follows Kya Clark, a young woman who grows up alone in the marshes of North Carolina. Owens crafts a tale of resilience, survival, and the transformative power of nature, entrancing readers with its evocative prose and mesmerizing setting. These bestselling novels represent just a fraction of the literary treasures that have emerged in 2023. Whether you seek tales of romance, adventure, or personal growth, the world of literature offers an abundance of captivating stories waiting to be discovered. The novel begins with Richard Papen, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a quiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts. The Secret History is an exceptional and suspenseful novel that will keep you wondering until the very end. The novel is a warning tale about the dangers of obsession and the power of evil.

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