

Unit 1: Fundamentals of Chemistry

Exercise Questions

Exercise Multiple Choice Question Answers

- Industrial chemistry deals with the manufacturing of compounds:**
(a) in the laboratory (b) on micro scale
(c) on commercial scale (d) on economic scale
- Which one of the following can be separated by physical means?**
(a) mixture (b) element (c) compound (d) radical
- The most abundant element occurring in the oceans is:**
(a) oxygen (b) hydrogen (c) nitrogen (d) silicon
- Which one of the following element is found in most abundance in the earth's crust?**
(a) oxygen (b) aluminum (c) silicon (d) iron
- The third abundant gas found in the atmosphere is:**
(a) carbon monoxide (b) oxygen (c) nitrogen (d) argon
- One amu (atomic mass unit) is equivalent to:**
(a) 1.66×10^{-24} mg (b) 1.66×10^{-24} g (c) 1.66×10^{-23} g (d) 1.66×10^{-23} g
- All of the followings are tri-atomic molecule except:**
(a) H_2 (b) O_3 (c) H_2O (d) CO_2
- The mass of one molecule of water is:**
(a) 18 amu (b) 18 g (c) 18 mg (d) 18 kg
- The molar mass of H_2SO_4 is:**
(a) 98 g (b) 48 amu (c) 4.8 g (d) 98 amu
- Molar mass is usually expressed in grams. Which one of the following is molar mass of O_2 in amu?**
(a) 32 amu (b) 53.2×10^{-24} amu
(c) 1.92×10^{-25} amu (d) 192.64×10^{-25} amu
- How many numbers of moles are equivalent to 8 grams of CO_2 ?**
(a) 0.15 (b) 0.18 (c) 0.21 (d) 0.24
- Which one of the following pairs has the same number of ions?**
(a) 1 mole of NaCl and 1 mole of $MgCl_2$
(b) 1/2 mole of NaCl and 1/2 mole of $MgCl_2$
(c) 1/2 mole of NaCl and 1/3 mole of $MgCl_2$
(d) 1/3 mole of NaCl and 1/2 mole of $MgCl_2$
- Which one of the following pairs has the same mass?**
(a) 1 mole of CO and 1 mole of N_2 (b) 1 mole of CO and 1 mole of CO_2
(c) 1 mole of O_2 and 1 mole of N_2 (d) 1 mole of O_2 and 1 mole of CO_2

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