

Unit 1: Fundamentals of Chemistry

Exercise Questions

Exercise Multiple Choice Question Answers

- Industrial chemistry deals with the manufacturing of compounds:**
(a) in the laboratory (b) on micro scale
(c) on commercial scale (d) on economic scale
- Which one of the following can be separated by physical means?**
(a) mixture (b) element (c) compound (d) radical
- The most abundant element occurring in the oceans is:**
(a) oxygen (b) hydrogen (c) nitrogen (d) silicon
- Which one of the following element is found in most abundance in the earth's crust?**
(a) oxygen (b) aluminum (c) silicon (d) iron
- The third abundant gas found in the atmosphere is:**
(a) carbon monoxide (b) oxygen (c) nitrogen (d) argon
- One amu (atomic mass unit) is equivalent to:**
(a) 1.66×10^{-24} mg (b) 1.66×10^{-24} g (c) 1.66×10^{-23} g (d) 1.66×10^{-23} g
- All of the followings are tri-atomic molecule except:**
(a) H_2 (b) O_3 (c) H_2O (d) CO_2
- The mass of one molecule of water is:**
(a) 18 amu (b) 18 g (c) 18 mg (d) 18 kg
- The molar mass of H_2SO_4 is:**
(a) 98 g (b) 48 amu (c) 4.8 g (d) 98 amu
- Molar mass is usually expressed in grams. Which one of the following is molar mass of O_2 in amu?**
(a) 32 amu (b) 53.2×10^{-24} amu
(c) 1.92×10^{-25} amu (d) 192.64×10^{-25} amu
- How many numbers of moles are equivalent to 8 grams of CO_2 ?**
(a) 0.15 (b) 0.18 (c) 0.21 (d) 0.24
- Which one of the following pairs has the same number of ions?**
(a) 1 mole of NaCl and 1 mole of $MgCl_2$
(b) 1/2 mole of NaCl and 1/2 mole of $MgCl_2$
(c) 1/2 mole of NaCl and 1/3 mole of $MgCl_2$
(d) 1/3 mole of NaCl and 1/2 mole of $MgCl_2$
- Which one of the following pairs has the same mass?**
(a) 1 mole of CO and 1 mole of N_2 (b) 1 mole of CO and 1 mole of CO_2
(c) 1 mole of O_2 and 1 mole of N_2 (d) 1 mole of O_2 and 1 mole of CO_2

Solved Exercises Of Chemistry Of 11th Hsc

Michael Rajnik



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Excel HSC Chemistry Jim Stamell, 2011 ISBN 9781741252996 AUTHOR Jim Stamell RRP 39 95 PAGES 428 pp SPECIFICATION Softcover perfect bound 280 mm x 210 mm STATUS New edition PUBLICATION DATE April 2008 The EXCEL HSC Chemistry guide is directly linked to the syllabus with every single dot point of the HSC Chemistry syllabus appearing in the margin of the book You can write in the guide so your study is focused and your notes are structured This guide comes in a brand new format that makes even better use of your study time up to date coverage of the core topics plus 3 Option topics Industrial Chemistry Shipwrecks Corrosion and Conservation and Forensic Chemistry this guide is organised just like the HSC syllabus so the students learn to section the theoretical part is under routine headings and the students section the practical part is under headings like First hand Second hand and Investigations and Problem Solving %this way you will be able to see at a glance what the theoretical and practical work is all main headings in each chapter 1 1 2 1 etc are directly from the syllabus word for word %this way you can easily match the Excel guide to the syllabus an alphabetical list of all the key definitions and concepts you should know from each chapter %an efficient way of learning all the definitions in one go chapter syllabus checklist with every single dot point listed in checklist form for each chapter %a fantastic way of testing that you know all the work hundreds of key concept questions with answers %questions that test you recall of knowledge in each chapter HSC type questions for every section in each chapter with clock icons to tell you how much time you will have to answer the questions in the HSC %this way you can test yourself on HSC type questions under HSC type time pressure an examiner maximiser feature ticks to show the mark distribution and answers to all HSC type questions %all you need to answer HSC type questions two sample HSC papers with an examiner maximiser feature plus answers %not one but two up to date sample papers the Excel syllabus summary notes a detachable section at the end of the guide where every single dot point of each chapter is summarised for you% a comprehensive and compact summary of the whole course in 32 pages

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