

10.1 Periodic Trends



10.1 PERIODIC TRENDS: FROM CONCEPT TO APPLICATION

Section 10.1: Introduction

In your notebook, read aloud the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

Atomic Mass	_____
Atomic Number	_____
Electron Configuration	_____
Chemical Name	_____
Chemical Symbol	_____

Put each item in Column A, either the letter of the matching item in Column B.

Column A

- _____ 10. It varies on the periodic table.
- _____ 11. It has no fixed periodic table.
- _____ 12. Group 1 elements.
- _____ 13. Elements that are strong oxidizing agents.
- _____ 14. Group 17 elements.

Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write one of the statements in terms of the statement in italics, change the underlined word or phrase to make it true.

- _____ 15. There are two representative elements of elements.
- _____ 16. Most elements have fixed of the elements in the periodic table are metals.
- _____ 17. Group 1A elements except for hydrogen are metals in the solid state.
- _____ 18. Group 2A elements are the alkaline earth metals.
- _____ 19. Group 1A elements are highly reactive elements known as alkali metals.
- _____ 20. Group 1A elements are very reactive elements known as representative elements.
- _____ 21. Alkali metals have properties of both metals and non-metals.

Study Guide For Content Mastery Chapter 1answer Key

Lei Huang



Study Guide For Content Mastery Chapter 1 answer Key:

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