

10.1 Periodic Trends



PERIODIC TRENDS: FROM CONCEPT TO PRACTICE

Section 10.1 (continued)

In your notebook, read aloud the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

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Atomic Mass	_____
Atomic Number	_____
Electron Configuration	_____
Chemical Name	_____
Chemical Symbol	_____

Put each item in Column A, either the letter of the matching item in Column B.

Column A

- _____ 1. It increases in the periodic table.
- _____ 2. It decreases in the periodic table.
- _____ 3. Group 1 elements.
- _____ 4. Elements that are shiny and conduct electricity.
- _____ 5. Group 17 elements.

Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write one of the statements in terms of the statement in italics, change the underlined word as shown in **Model 10.1**.

- _____ 1. *There are two representative elements of elements.*
- _____ 2. *There are three periods of the elements in the periodic table.*
- _____ 3. *Group 1 elements include hydrogen and lithium in the alkali metals.*
- _____ 4. *Group 17 elements are the halogens with iodine.*
- _____ 5. *Group 1 elements are highly reactive elements known as alkalis.*
- _____ 6. *Group 17 elements are very reactive elements known as halogens.*
- _____ 7. *Alkaloids have properties of both metals and non-metals.*

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David Baud



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