

CUET Chemistry MCQs

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Solutions

- Which of the following is not a binary solution?
 (A) Pure water + Sugar
 (B) Air
 (C) Mixture of benzene and toluene
 (D) Mixture of ethanol and methanol
- Which of the following is not an example of saturated solution?
 (A) Sugar syrup
 (B) Salt water
 (C) Honey
 (D) Carbonated drink
- It sea water is assumed to contain 40g of salt per 200g of sea water then percentage by mass of salt present is
 (A) 15%
 (B) 40%
 (C) 20%
 (D) 10%
- If 2 moles of CaCO_3 is dissolved in 900g of water then percentage by mass of solution formed is
 (A) 36.36%
 (B) 9.9%
 (C) 18.18%
 (D) 11.75%
- 0.5 moles of benzene was dissolved in 2 moles of toluene. Find mass % of benzene in the solution formed
 (A) 25.48%
 (B) 9.52%
 (C) 15.92%
 (D) 17.48%
- What is the number of ppm of NaCl in the solution with 117g of NaCl dissolved in 500 mL of water?
 (A) 1.89×10^5
 (B) 2.78×10^5
 (C) 9.12×10^5
 (D) 6.54×10^5
- What is the concentration of mercury in ppm if 3g mercury is found in 15kg sample of ocean water?
 (A) 300
 (B) 400
 (C) 500
 (D) 200
- If 3.0 mg of Cr^{+6} is dissolved in 3 L of water, find the concentration of Cr^{+6} in ppm.
 (A) 2 ppm
 (B) 1 ppm
 (C) 3 ppm
 (D) 9 ppm
- What is the number of ppm of methanol dissolved in 68g of water (Taking mass of methanol 32gm)
 (A) 160000
 (B) 480000
 (C) 320000
 (D) 520000
- A tank contains 5 moles of oxygen, 2 moles of nitrogen and 20g of hydrogen at room temperature. Find the mole fraction of hydrogen.
 (A) $\frac{5}{17}$
 (B) $\frac{2}{15}$
 (C) $\frac{8}{15}$
 (D) $\frac{10}{17}$
- Which of the following is insoluble in water?
 (A) salt
 (B) acid
 (C) base
 (D) oil
- On increasing temperature, the solubility of NaOH in water.
 (A) Increases
 (B) Decreases
 (C) Remains unaffected
 (D) Cannot be determined
- On increasing temperature, the solubility of NaNO_3 in water.
 (A) Increases
 (B) Decreases
 (C) Remains unaffected
 (D) Cannot be determined
- What is the effect of complex formation on solubility of I_2 ?
 (A) Increases
 (B) Decreases
 (C) Remains same
 (D) None of the above
- Solid forms of carboxylic acids are soluble in
 (A) Water
 (B) Acetone
 (C) Wax
 (D) Oil
- Fishes feel uncomfortable in warm water due to
 (A) fishes do not like warmth
 (B) higher amount of impurities

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