

2-7 Reteaching

Solving Proportions

A proportion is an equation that states that two ratios are equal. If a quantity in a proportion is unknown, you can solve a proportion to find the unknown quantity as shown below.

Problem

What is the solution of $\frac{3}{4} = \frac{x}{14}$?

There are two methods for solving proportions—using the Multiplication Property of Equality and the Cross Products Property.

- 1) The Multiplication Property of Equality says that you can multiply both sides of an equation by the same number without changing the value.

$$\begin{aligned}\frac{3}{4} &= \frac{x}{14} \\ 14\left(\frac{3}{4}\right) &= 14\left(\frac{x}{14}\right) \\ \frac{42}{4} &= x \\ 10.5 &= x\end{aligned}$$

To isolate x , multiply each side by 14.

Simplify.

Divide 42 by 4.

- 2) The Cross Products Property says that you can multiply diagonally across the proportion and these products are equal.

$$\begin{aligned}\frac{3}{4} &= \frac{x}{14} \\ (4)(x) &= (3)(14) \\ 4x &= 42 \\ \frac{4x}{4} &= \frac{42}{4} \\ x &= 10.5\end{aligned}$$

Multiply diagonally across the proportion.

Multiply.

To isolate x , divide each side by 4.

Simplify.

Real world situations can be modeled using proportions.

Problem

A bakery can make 6 dozen donuts every 21 minutes. How many donuts can the bakery make in 2 hours?

A proportion can be used to answer this question. It is key for you to set up the proportion with matching units in both numerators and both denominators.

For this problem, you know that 2 hours is 120 minutes and 6 dozen is 72 donuts.

Correct:

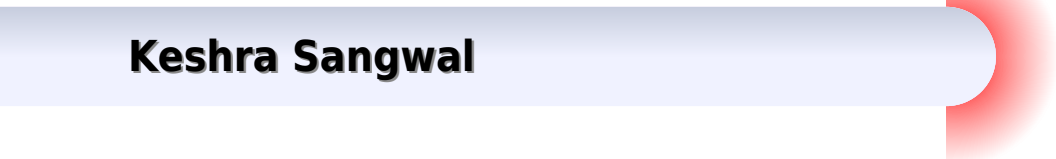
$$\frac{72 \text{ donuts}}{21 \text{ min}} = \frac{x \text{ donuts}}{120 \text{ min}}$$

Incorrect:

$$\frac{72 \text{ donuts}}{21 \text{ min}} = \frac{120 \text{ min}}{x \text{ donuts}}$$

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Keshra Sangwal



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Prentice Hall Algebra 1, 1990 **Prentice Hall Geometry** Robert Kalin, Mary Kay Corbitt, 1990 **Solutions Manual** Henry Stark, John Woods, Jr., 2001-10 Thermodynamics of Solutions Eli Ruckenstein, Ivan L. Shulgin, 2009-06-17 This book consists of a number of papers regarding the thermodynamics and structure of multicomponent systems that we have published during the last decade. Even though they involve different topics and different systems, they have something in common which can be considered as the signature of the present book. First, these papers are concerned with difficult or very nonideal systems, i.e. systems with very strong interactions, e.g. hydrogen bonding between components or systems with large differences in the partial molar volumes of the components, e.g. the aqueous solutions of proteins or systems that are far from normal conditions, e.g. critical or near critical mixtures. Second, the conventional thermodynamic methods are not sufficient for the accurate treatment of these mixtures. Last but not least, these systems are of interest for the pharmaceutical, biomedical, and related industries. In order to meet the thermodynamic challenges involved in these complex mixtures, we employed a variety of traditional methods but also new methods such as the fluctuation theory of Kirkwood and Buff and ab initio quantum mechanical techniques. The Kirkwood-Buff (KB) theory is a rigorous formalism which is free of any of the approximations usually used in the thermodynamic treatment of multicomponent systems. This theory appears to be very fruitful when applied to the above-mentioned difficult systems.

Prentice Hall Algebra Prentice Hall (School Division), **CRC Handbook of Solubility Parameters and Other Cohesion Parameters** Allan F.M. Barton, 2017-10-19 The CRC Handbook of Solubility Parameters and Other Cohesion Parameters, Second Edition, which includes 17 new sections and 40 new data tables, incorporates information from a vast amount of material published over the last ten years. The volume is based on a bibliography of 2,900 reports, including 1,200 new citations. The detailed, careful construction of the handbook develops the concept of solubility parameters from empirical thermodynamic and molecular points of view and demonstrates their application to liquid, gas, solid, and polymer systems.

The Solutions Focus Paul Jackson, Mark McKergow, 2011-02-22 The revolutionary yet radical alternative, the solutions-focused approach to discovering what works at work.

Organized Solutions Stig Friberg, 1992-07-21 Written by top international experts in colloid and surface chemistry. Contains close to 750 literature references and nearly 400 useful figures, equations, and tables.

Solvents and Solutions: Structure and Properties Keshra Sangwal, 2021-08-06 A UNIQUE BOOK ON THE PRESENT STATUS OF SOLVENTS AND SOLUTIONS WITH IMPORTANT PROBLEMS RELATED TO THEIR STRUCTURE AND PROPERTIES. The literature on the properties of solvents and solutions used in academic research and in a wide range of industries has grown enormously during the last four decades and is scattered in different specialized journals. *Solvents and Solutions* is a groundbreaking text that offers a systematic compilation of important problems related to selected properties of solvents and solutions based on the literature published so far. The author places emphasis on explaining the basic concepts involved in understanding the properties and

behavior of various solvents and solutions of electrolytes and nonelectrolytes in a consistent manner After a description of the general characteristics of structure of solvents and solutions and the solubility of electrolytes and nonelectrolytes under normal temperature and pressure conditions the book first deals with different aspects of the density and the refractive index of solvents and dilute as well as concentrated solutions and finally with the transport i e viscosity and electric conductivity and thermal properties of solvents and solutions Solvents and solutions is the first text devoted to the description and discussion of their properties since the publication of a monograph on the physical properties of aqueous electrolyte solutions more than three decades ago The main features of this book are Reflects developments in the investigation of solvents and solutions during the last three decades Outlines basic concepts involved in understanding the properties and behavior of solvents and solutions Describes and discusses different properties of ionic liquids as solvents and the behavior of their mixtures with other commonly used solvents Contents of different chapters are not only self contained but the contents are practically independent of each other Written as a practical guide for researchers who are looking for an uptodate overview of the physical and transport properties of solvents and solutions and as a reference source for workers in chemical industries and related fields and for graduate students of chemical engineering and physical chemistry *Prentice Hall Algebra 2* Prentice Hall (School Division), Solution Behavior of Surfactants K.L. Mittal,E.J. Fendler,2012-12-06 This and its companion Volume 2 comprise the proceedings of the International Symposium on Solution Behavior of Surfactants Theoretical and Applied Aspects organized under the auspices of the 11th Northeast Regional Meeting of the American Chemical Society held in Potsdam N Y June 30 July 3 1980 This Symposium represented the third event in the series of symposia dealing with the topic of surfactants in solution The first Symposium was held in Albany N Y in 1976 under the title Micellization Solubilization and Microemulsions 1 the proceedings of which have been documented in a two volume set The second was held under the title Solution Chemistry of Surfactants in 1978 in Knoxville TN and the proceedings of this event have also been properly chronicled Apropos the fourth biennial Symposium in this series is entitled International Symposium on Surfactants in Solution K L Mittal and B Lindman Cochairmen and is scheduled to be held from June 27 to July 2 1982 in Lund Sweden Since these biennial events have been very successful and important in bringing researchers with varied interests together and in stimulating interdisciplinary communication so the plans are to continue these on a regular basis with a change in venue for each meeting **Prentice-Hall Federal Income Tax Course** Prentice-Hall, Inc,1925

Standard Potentials in Aqueous Solution Allen J. Bard,Roger Parsons,Joseph Jordan,2017-11-22 The best available collection of thermodynamic data The first of its kind in over thirty years this up to date book presents the current knowledge on Standard Potentials in Aqueous Solution Written by leading international experts and initiated by the IUPAC Commissions on Electrochemistry and Electroanalytical Chemistry this remarkable work begins with a thorough review of basic concepts and methods for determining standard electrode potentials Building upon this solid foundation this convenient

source proceeds to discuss the various redox couples for every known element. The chapters of this practical time saving guide are organized in order of the groups of elements on the periodic table for easy reference to vital material AND each chapter also contains the fundamental chemistry of elements, numerous equations of chemical reactions, easy to read tables of thermodynamic data and useful oxidation state diagrams. Standard Potentials in Aqueous Solution is an ideal handy reference for analytical and physical chemists, electrochemists, electroanalytical chemists, chemical engineers, biochemists, inorganic and organic chemists and spectroscopists needing information on reactions and thermodynamic data in inorganic chemistry. And it is a valuable supplementary text for undergraduate and graduate level chemistry students. **Solution-Focused Brief**

Therapy Cynthia Franklin, 2012. Therapy is frequently miscast as requiring an enormous amount of time and financial commitment, but helpful goal oriented therapy can produce positive results after only a few sessions. By focusing on solutions instead of problems, SFBT asks clients to set concrete goals and to draw upon strengths in their lives that can help bring about the desired change for a preferred future. *Solutions Manual* Tirupathi Chandrupatla, Belegunda, 2002-02

Prentice Hall Trigonometry Jerome D. Hayden, Bettye C. Hall, 1990 *Selected Solutions Manual for Principles of Chemistry* Mary Beth Kramer, Nivaldo J. Tro, Kathleen Thrush Shaginaw, 2009-06-01 **NASA Technical Note**, 1964

Lasers in Polymer Science and Technology Jan F. Rabek, Jean-Pierre Fouassier, 1989-11-30. The purpose of this 4 volume book is to examine some of the applications of lasers in polymer science and technology. Now available for the first time, up to date information on this fascinating subject is compiled and presented in compact form. This book focuses on current research and developments in the application of lasers in polymer and biopolymer chemistry. It includes experimental and theoretical details, apparatus, techniques and applications. This book is a useful source for researchers, students, polymer chemists and physicists involved in this astonishing field of high technology. **Use of Gas Diffusion Membrane**

Electrodes to Investigate Coordination Species in the Sulfur Dioxide System S. E. Khalafalla, Richard H. Jefferson, 1978

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