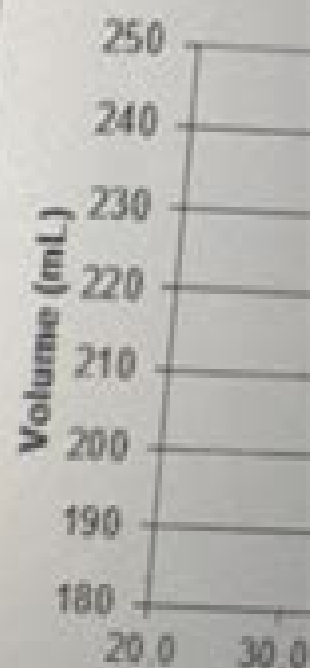


## Gas Laws

## Experiment 2 - Charles' Law

| Dial<br>setting | Pressure<br>P, (torr) | Temp.<br>T, (°C) | Temp.<br>T, (K) | Height<br>(cm) | Volume<br>V, (mL) | Constant? |
|-----------------|-----------------------|------------------|-----------------|----------------|-------------------|-----------|
| 0               |                       |                  |                 |                |                   |           |
| 1               |                       |                  |                 |                |                   |           |
| 2               |                       |                  |                 |                |                   |           |
| 3               |                       |                  |                 |                |                   |           |
| 4               |                       |                  |                 |                |                   |           |
| 5               |                       |                  |                 |                |                   |           |
| 6               |                       |                  |                 |                |                   |           |
| 7               |                       |                  |                 |                |                   |           |

Note: As you fill in the height and Celsius temperature values, the Kelvin temperature and the volume of the cylinder will be calculated for you.



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