

**CHEMISTRY**  
**PAPER – 2**  
**(PRACTICAL)**

*(Maximum Marks: 30)*

*(Time allowed: Three hours)*

*(Candidates are allowed additional 15 minutes for only reading the paper.  
They must NOT start writing during this time.)*

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**ALL ANSWERS MUST BE WRITTEN IN THE ANSWER BOOKLET PROVIDED  
SEPARATELY.**

*Question 1 is an oxidation-reduction titration in which sufficient working details are given.  
All essential working must be shown.*

*Question 2 is an exercise dealing with identification of organic compounds.  
Credit will be given for precise observations recorded and for well-drawn deductions.*

*Question 3 is an exercise in qualitative analysis.*

---

*Read the questions carefully and follow the given instructions.*

*Attempt all questions.*

*All working, including rough work, should be done on the same sheet as the  
rest of the answer.*

*The intended marks for questions or parts of questions are given in brackets [ ].*

*Mathematical Tables are provided.*

---

*Attempt all questions.*

**Question 1**

**[7]**

You are provided with two solutions as follows:

- C-10 is a solution containing 1.95 gms of potassium manganate (VII),  $\text{KMnO}_4$  per litre.
  - C-11 is a solution prepared by dissolving 23.2 gms of hydrated ammonium iron (II) sulphate crystals,  $(\text{NH}_4)_2\text{SO}_4 \cdot \text{FeSO}_4 \cdot x\text{H}_2\text{O}$  per litre.
- 

**This Paper consists of 4 printed pages.**

**Turn over**

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# Waec Chemistry Practical Question Paper 24

**Graham Chadwick Hill**



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**Chemistry Practical** Cape Technikon School of Physical Sciences, 198?      **Cambridge International AS and a Level Chemistry Practical Skills Workbook** Olli Dunning, Judy Potter, 2020-10-30 This title has been endorsed by Cambridge Assessment International Education for examination to support the full syllabus for examination from 2022 Reinforce learning and deepen understanding of the practical skills required by the revised syllabus ideal as course companions or homework books for use when carrying out and analysing practical work throughout the course Support students learning and provide guidance on practical skills with extra practice questions and activities tailored to topics in the Student Book Keep track of students work with ready to go write in exercises which once completed can also be used to recap learning for revision Offer extra support for the mathematical and statistical parts of the course Answers can be found at [www.hoddereducation.com/cambridgeextras](http://www.hoddereducation.com/cambridgeextras)      **Key to Chemistry** Graham Chadwick Hill, 1978      **Keys to Chemistry.**, 1978

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and precision Students navigate the intricacies of solubility rules equipping them with the knowledge to predict the outcomes of various combinations of substances The book fosters an intuitive grasp of solubility principles ensuring students approach experiments with a solid foundation Flame tests add a vibrant and visually appealing dimension to the practical chemistry experience With colorful illustrations and insightful explanations the book guides students through the identification of elements based on the characteristic colors they emit when heated This section not only enhances practical skills but also sparks curiosity and fascination for the visual aspects of chemistry Beyond the individual topics the book addresses the overarching theme of success in the KCSE Chemistry Practical exam It goes beyond rote memorization emphasizing a holistic approach to learning Practical demonstrations are woven seamlessly into the fabric of the content encouraging students to actively engage with the material This hands on approach not only aids in exam preparation but also fosters a genuine love for the subject A dedicated section on exam strategies serves as a valuable toolkit for students From time management tips to effective approaches for different question types the book ensures students enter the exam hall well prepared and confident The friendly tone and approachable style of the book create a supportive learning environment fostering a positive mindset crucial for success How to Pass KCSE Chemistry Practical is not just a guide it is a mentor that understands the challenges students face and provides the necessary support for overcoming them The author s extensive teaching experience shines through offering a level of insight and guidance that goes beyond the typical textbook This book is an invitation to explore the wonders of chemistry encouraging students to view the subject not as a series of tasks to be completed but as a lifelong journey of discovery Chemistry (Practical Examinations)., 1932

**Chemistry Questions Theoretical and Practical. Selected from the Papers of the Civil Service Commission** Arthur Percival Newton, 1912

**Chemistry (Practical A).**, 1946 **Chemistry** Reger, 1993 **Comprehensive Practical Chemistry (Abhilekhan)**  
**XII** Dr. N.K. Verma, Avinash Ojha, 2008-05-01 Practical Skills, 2008 **Papers in Practical Chemistry** Allen  
 Rogers, 1902 **Chemistry Practicals** K. N. Rastogi, 2010 *Comprehensive Chemistry Practical Notebook Xii* Dr. N. K.  
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