- 16. Increasing the temperature generally\_\_\_\_
  - decreases the solubility of a solid in a liquid but increases the solubility of a gas in a liquid.
  - increases the solubility of a solid in a liquid but decreases the solubility of a gas in a liquid.
  - Increases the solubility of both a solid and a gas in a liquid.
  - D. decreases the solubility of both a solid and a gas in a liquid.
- A white precipitate was formed when BaCl<sub>2(aq)</sub> was added to an aqueous solution of a salt X. The precipitate dissolved in dilute HCl with rapid efferyescence. Salt X is
  - likely to contain

    A. SO<sub>4</sub><sup>2-</sup> ions.
  - B. SO<sub>3</sub><sup>2</sup>- ions.
  - C. NO<sub>3</sub> ions.
  - D. CO<sub>3</sub><sup>2-</sup> ions.
- Before a reaction could take place, there should be
  - A. ionization of reactant particles.
  - B. breakage of bonds of products.
  - C. breakage of bonds of reactants.
  - D. ionization of product particles.
- 19. Consider the following reaction equation:

Silicon(IV) exide is acting as

- A. a basic oxide.
- B. a reducing agent.
- an acidic oxide.
- D. an oxidizing agent.

- 20. Which of the following acids would form normal salt only?
  - A. Tetraoxosulphate(VI) acid
  - B. Tricoconitrate(V) acid
  - C. Tetraoxophosphate(V) acid
  - D. Trioxosulphate(IV) acid
  - What is the partial pressure of oxygen at s.t.p.
    in a gaseous mixture containing 100 cm<sup>3</sup>
    of oxygen and 900 cm<sup>3</sup> of nitrogen gas?
    - A. 1.0 arm
    - B. 0.1 artor
    - C. 9.0 atm
    - D. 0.9 over
  - Graphite is used as a dry lubricant due to the presence of
    - A. mobile electrons.
    - B. free electrons.
    - C. octahedral structures.
    - D. layered structures.
- 23. Which of the following gases is alkaline?
  - A. NO.
  - B. CO<sub>2</sub>
  - C. NH<sub>3</sub>
  - D. N<sub>2</sub>O
- 24. Which of the following statements about an equilibrium system is correct?
  - Forward and backward reactions occur at the same rate.
  - The concentrations of reactants must equal that of the products.
  - The concentrations of reactants and products can be changed by adding a catalyst.
  - D. The concentrations of reactants and products are not affected by a change in temperature.

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