

# WHALE TALK



By the author of *Stargazing for Fun & Profit*, *Pyro*

CHRIS CRUTCHER

# Whale Talk Chris Crutcher

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web may 15 2020 toluene 1 oxidation with  $\text{KMnO}_4$  mechanism 2 benzoic acid  $\text{C}_6\text{H}_5\text{COOH}$   $\text{SOCl}_2$  3 benzoyl chloride  $\text{C}_6\text{H}_5\text{COCl}$   $\text{LiAlH}_4$   $\text{CH}_3\text{CH}_2\text{CH}_2\text{CHO}$  4 benzaldehyde  $\text{C}_6\text{H}_5\text{CHO}$   $\text{CH}_2\text{OHCH}_2\text{OH}$  h step 1 toluene oxidation with  $\text{KMnO}_4$   $\text{KMnO}_4$  is a strong oxidising agent which oxidise toluene to benzaldehyde in 1st step

**oxidation of alkylarenes to the corresponding acids using** - Dec 07 2022

web may 1 2004 oxidation of toluene using aqueous potassium permanganate was studied under heterogeneous condition in the presence of hydrodynamic cavitation and compared with the results of the reaction under acoustic cavitation

**can acidified or neutral  $\text{KMnO}_4$  oxidise toluene to benzoic acid** - Jan 08 2023

web oct 15 2014 1 answer sorted by 6 here are the three equations describing the reduction of manganese and concurrent oxidation of whatever substrate may be present under basic neutral and acidic conditions respectively  $\text{MnO}_4^- + 7\text{H}^+ + 5\text{e}^- \rightarrow \text{Mn}^{2+} + 4\text{H}_2\text{O}$   $\text{MnO}_4^- + 4\text{H}^+ + 3\text{e}^- \rightarrow \text{MnO}_2 + 2\text{H}_2\text{O}$   $\text{MnO}_4^- + 2\text{H}_2\text{O} + 3\text{e}^- \rightarrow \text{MnO}_2 + 4\text{OH}^-$  basic  $\text{MnO}_4^- + 7\text{H}^+ + 5\text{e}^- \rightarrow \text{Mn}^{2+} + 4\text{H}_2\text{O}$   $\text{MnO}_4^- + 4\text{H}^+ + 3\text{e}^- \rightarrow \text{MnO}_2 + 2\text{H}_2\text{O}$   $\text{MnO}_4^- + 2\text{H}_2\text{O} + 3\text{e}^- \rightarrow \text{MnO}_2 + 4\text{OH}^-$  basic

**what is the mechanism of toluene oxidation by  $\text{KMnO}_4$  reddit** - Jul 14 2023

web mar 15 2019 what is the mechanism of toluene oxidation by  $\text{KMnO}_4$  it is quite simple reaction that  $\text{KMnO}_4$  can oxidize benzylic hydrogen in toluene and synthesize benzoic acid however i cannot find exact mechanism of reaction in

**oxidation of aromatic alkanes with  $\text{KMnO}_4$  to give carboxylic acids** - Feb 09 2023

web description treatment of an alkylbenzene with potassium permanganate results in oxidation to give the benzoic acid notes the position directly adjacent to an aromatic group is called the benzylic position the reaction only works if there is a hydrogen attached to the carbon examples

**oxidation of alkylarenes to the corresponding acids using** - Mar 10 2023

web may 1 2004 the oxidation of toluene by aq  $\text{KMnO}_4$  under hydrodynamic cavitation was taken as a model reaction and various parameters have been optimized the oxidation of toluene by aq  $\text{KMnO}_4$  gives benzoic acid scheme 1  $\text{KMnO}_4$  in turn is reduced to  $\text{MnO}_2$  the reaction did not give any other byproduct