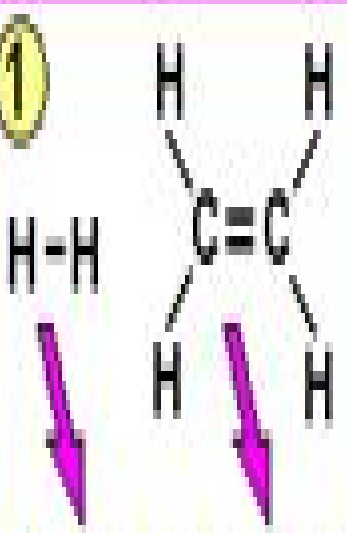
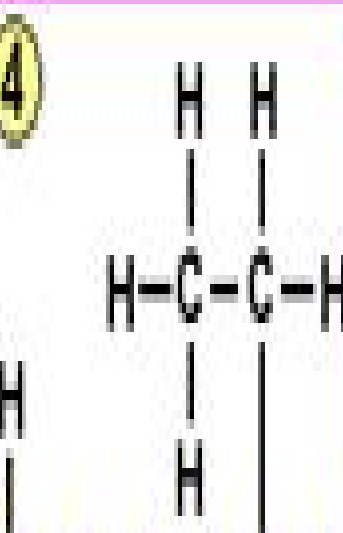
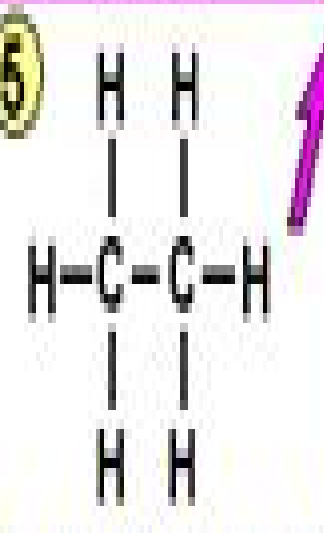


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the hydrogen and ethene molecules collide with the nickel catalyst surface (c) doc b	both reactant molecules become physically adsorbed on to the catalyst surface, held by inter-molecular forces	the H-H and 'half' of the C=C bond are broken and the 'spare' H and C electrons form bonds with the nickel, this is called chemisorption	Ni-H and Ni-C bonds break and a C-H bond is formed and is repeated to form the ethane molecule	the ethane molecule is temporarily held by intermolecular forces but will readily break free of the catalyst surface

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imminent researchers in science and mathematics teaching and learning Professor Alan Schonfeld from UC Berkeley USA and Professor Ilka Parchman from IPN at the University of Kiel Germany The book as a whole as well as its individual chapters are intended for a wide audience of curriculum developers teacher educators researchers on learning and teaching of science and mathematics and policy makers at the university level interested in advancing models of academic departments working under a common philosophy yet under full academic freedom Contributors are Abraham Arcavi Michal Armoni Ron Blonder Miriam Carmeli Jason Cooper Rachel Rosanne Eidelman Ruhama Even Bat Sheva Eylon Alex Friedlander Nurit Hadas Rina Hershkowitz Avi Hofstein Ronnie Karsenty Boris Koichu Dorothy Langley Ohad Levkovich Smadar Levy Rachel Mamlok Naaman Nir Orion Zahava Scherz Alan Schoenfeld Yael Shwartz Michal Tabach Anat Yarden and Edit Yerushalmi

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