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Study Guide to Accompany Chemical Principles, Properties, and Reactions Kenneth L. Henold, 1984 *Robinson Chemistry Study Guide* Robinson, 1992 Fundamentals of Chemistry, Study Guide James E. Brady, John R.

Holum,1988-04-20 This Third Edition of the widely used fundamentals textbook for science majors maintains the conversational writing style that made the previous editions so popular while including up to date treatments of important and current topics Emphasizes descriptive chemistry chemical reactions and properties while maintaining a solid treatment of chemical principles Common chemicals are used whenever possible as examples in both theoretical discussions and in problems and exercises Incorporates many pedagogical aids each chapter begins with a brief table of contents and each section begins with a preview of topics covered Chapters include frequent margin comments figures and photographs

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nuclear reactions all equations need to be balanced to conform to two conservation laws the mass number is conserved and the electrical charge of the nucleus is conserved