

Thermodynamic Feasible Reactions: (2000-2019)

10. Calculate the standard enthalpy of formation for the reaction



with $\Delta_f H^\circ$



with $\Delta_f H^\circ$ (2000-2019)

(2000-2019)

11. Calculate the free enthalpy of the combustion of vinyl acetate, with



with $\Delta_f G^\circ$



with $\Delta_f G^\circ$ (2000-2019)

with $\Delta_f G^\circ$ (2000-2019)

with $\Delta_f G^\circ$ (2000-2019)

(2000-2019)

12. Calculate the total required for thermodynamic stability, CaCO_3 , relative CaO , and natural elements, with all natural elements at standard conditions



with $\Delta_f G^\circ$ (2000-2019)

with $\Delta_f G^\circ$ (2000-2019)

with $\Delta_f G^\circ$ (2000-2019)

(2000-2019)

13. $\text{CH}_4 + 2\text{O}_2 \rightarrow \text{CO}_2 + 2\text{H}_2\text{O}$ with $\Delta_f H^\circ$ (2000-2019)



with $\Delta_f H^\circ$ (2000-2019)

with $\Delta_f H^\circ$ (2000-2019)

with $\Delta_f H^\circ$ (2000-2019)

(2000-2019)

14. Organometallics: Calculate the average bond that is considered a weak as is known. Calculate the standard enthalpy of formation for organometallics, given the following standard enthalpies of reaction



with $\Delta_f H^\circ$ (2000-2019)

with $\Delta_f H^\circ$ (2000-2019)

with $\Delta_f H^\circ$ (2000-2019)

(2000-2019)

15. Two thermodynamic stability changes is related to the thermodynamic of systems



Calculate the standard free enthalpy of formation for organometallics, given the following standard enthalpies of formation: $\Delta_f H^\circ_{\text{BaSO}_4} = -1075 \text{ kJ mol}^{-1}$ with $\Delta_f H^\circ_{\text{HClO}_4} = -103 \text{ kJ mol}^{-1}$ (2000-2019)

(2000-2019)

16. Two thermodynamic stability changes is related to the thermodynamic of systems

with $\Delta_f H^\circ$ (2000-2019) (2000-2019)

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J Dewey



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