CHEMISTRY 11 – WORD EQUATIONS AND TYPES OF CHEMICAL REACTIONS WORKSHEET

- Translate the following word equations into chemical symbols and balance the resulting equations.
 Do not include the phases.
 - (a) Potassium + water → potassium hydroxide + hydrogen

(b) Nitrogen + hydrogen → ammonia

- In each of the following write a balanced chemical equation, including the phases.
 - (a) Liquid phosphoric acid reacts with aqueous barium hydroxide to give water and a precipitate of barium phosphate

(b) Solid aluminium oxide and aqueous sulphuric acid produce water and aqueous aluminium sulphate

(c) Nitrogen trifluoride gas and hydrogen gas react to form nitrogen gas and gaseous hydrogen fluoride

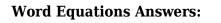
(d) Powdered sodium carbonate and aqueous hydrobromic acid react to form carbon dioxide gas, aqueous sodium bromide and water

Classify the following reactions.

Reaction	Туре
2 C ₂ H ₂ + 5 O ₂ → 4 CO ₂ + 2 H ₂ O	Combustion
Mg + CuSO ₄ → MgSO ₄ + Cu	Single Replacement
4 Na + O₂ → 2 Na₂O	Synthesis
$Sn(OH)_4 + 4 HBr \rightarrow 4 H_2O + SnBr_4$	Neutralization
2 Fe(NO ₃) ₃ + 3 MgS → Fe ₂ S ₃ + 3 Mg(NO ₃) ₂	Double Replacement
$2 N_2O \rightarrow 2 N_2 + O_2$	Decomposition

Word Equations Answers

David Baker



Reviewing Word Equations Answers: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is really astonishing. Within the pages of "**Word Equations Answers**," an enthralling opus penned by a highly acclaimed wordsmith, readers set about an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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