

Name: _____ Date: _____ Period: _____

Atoms and Molecules Worksheet

Using the Periodic Table located inside your text book, answer the following questions.

Analyzing Data: The information below is from part of the Periodic Table of Elements. Use this information to answer the questions that follow.

IA							VIII
1 H 1							2 He 4
	IIA	IIIB	IVB	VB	VIB	VIIIB	
3 Li 7	4 Be 9	5 B 11	6 C 12	7 N 14	8 O 16	9 F 19	10 Ne 20
11 Na 23	12 Mg 24	13 Al 27	14 Si 28	15 P 31	16 S 32	17 Cl 35	18 Ar 40

1) How many protons, neutrons, and electrons are in each of the following atoms?

Lithium (Li)

p = _____

n = _____

e = _____

Magnesium (Mg)

p = _____

n = _____

e = _____

Carbon (C)

p = _____

n = _____

e = _____

Oxygen (O)

p = _____

n = _____

e = _____

Chlorine (Cl)

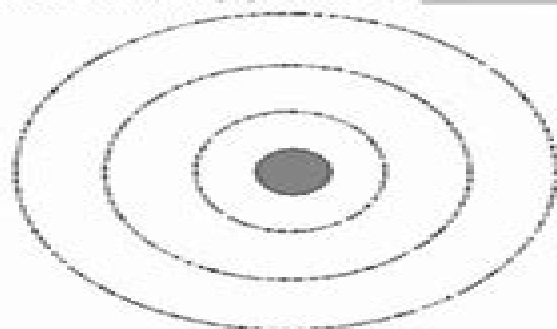
p = _____

n = _____

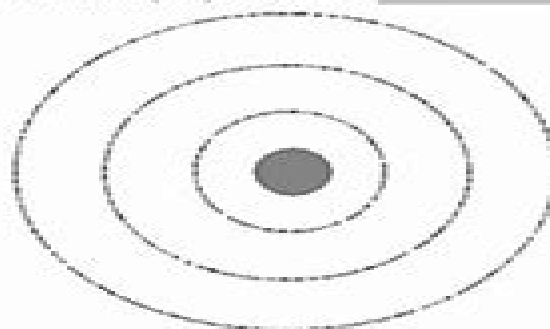
e = _____

2) Fill in the number of electrons in the correct energy levels below.

Number of Fluorine (F) electrons: _____



Number of Neon (Ne) electrons: _____



Is Fluorine Stable? _____

Explain your answer: _____

Is Neon Stable? _____

Explain your answer: _____

Worksheet 1 Rearranging Atoms

RM Cervero



Worksheet 1 Rearranging Atoms:

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