

Equilibrium Constant, K_{eq}

- Once a system has reached equilibrium, are the following true or false?
 - The reaction is finished, no more products are forming. false
 - The concentrations of the reactants and the products are equal. false
 - The concentrations are no longer changing. false
 - The reaction is not over, but will continue forever if isolated. true
 - The speed at which products are made equals the speed at which reactants form. true
- What is equal at equilibrium? rate forward = rate reverse
- What general information can be gathered by observing the magnitude of the equilibrium constant?

Whether the reactants or products are favoured.

- Write the expression for K_{eq} for the reaction: $2 \text{NO (g)} + \text{Cl}_2 \text{(g)} \leftrightarrow 2 \text{NOCl (g)}$

$$K_{eq} = \frac{[\text{NOCl}]^2}{[\text{NO}]^2 [\text{Cl}_2]}$$

- Write the K_{eq} for: $2 \text{K}_3\text{PO}_4 \text{(aq)} + 3 \text{Ca(NO}_3)_2 \text{(aq)} \leftrightarrow 6 \text{KNO}_3 \text{(aq)} + \text{Ca}_3(\text{PO}_4)_2 \text{(s)}$

$$K_{eq} = \frac{[\text{KNO}_3]^6}{[\text{K}_3\text{PO}_4]^2 [\text{Ca(NO}_3)_2]^3}$$

- Write the expression for K_{eq} for the reaction: $\text{H}_2 \text{(g)} + \text{Br}_2 \text{(l)} \leftrightarrow 2 \text{HBr (g)}$

$$K_{eq} = \frac{[\text{HBr}]^2}{[\text{H}_2]}$$

- Write the expression for K_{eq} for the reaction: $\text{CO}_2 \text{(g)} + \text{CaO (s)} \leftrightarrow \text{CaCO}_3 \text{(s)}$

$$K_{eq} = \frac{1}{[\text{CO}_2]}$$

- For the reaction: $\text{SiH}_4 \text{(g)} + 2 \text{O}_2 \text{(g)} \leftrightarrow \text{SiO}_2 \text{(g)} + 2 \text{H}_2\text{O (l)}$

a) Write the equilibrium expression for the forward reaction:

$$K_{eq} = \frac{[\text{SiO}_2]}{[\text{SiH}_4][\text{O}_2]^2}$$

b) Write the equilibrium expression for the reverse reaction:

$$K_{eq}' = \frac{[\text{SiH}_4][\text{O}_2]^2}{[\text{SiO}_2]} = \frac{1}{K_{eq}}$$

Worksheet 69 Equilibrium Constants

Karin Nielsen-Saines



Worksheet 69 Equilibrium Constants:

High Temperature Air Combustion Hiroshi Tsuji, Ashwani K. Gupta, Toshiaki Hasegawa, Masashi Katsuki, Ken Kishimoto, Mitsunobu Morita, 2002-12-03 Maximize efficiency and minimize pollution the breakthrough technology of high temperature air combustion HiTAC holds the potential to overcome the limitations of conventional combustion and allow engineers to finally meet this long standing imperative Research has shown that HiTAC technology can provide simultaneous reduction of CO₂ and nitric oxide emissions and reduce energy consumption for a specific process or requirement High Temperature Air Combustion From Energy Conservation to Pollution Reduction provides the first comprehensive exposition of the principles and practice of HiTAC With a careful balance of theory and practice it reviews the historical background clearly describes HiTAC combustion phenomena and shows how to simulate and apply the technology for significant energy savings reduced equipment size and lower emissions It offers design guidelines for high performance industrial furnaces presents field trials of practical furnaces and explores potential applications of HiTAC in other fields including the conversion of solid waste fuels to cleaner fuels stationary gas turbine engines internal combustion engines and other advanced energy to power conversion systems Developed through an intensive research project sponsored by the Japanese government HiTAC now promises to revolutionize our paradigm for using all kinds of fossil alternative waste and derived fuels for energy conversion and utilization in industry This book is your opportunity to understand its principles learn about the technology and begin to use it to the benefit of your application your company and the environment

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Financial Mathematics For Actuarial Science Richard James Wilders, 2020-01-24 Financial Mathematics for Actuarial Science The Theory of Interest is concerned with the measurement of interest and the various ways interest affects what is often called the time value of money TVM Interest is most simply defined as the compensation that a borrower pays to a lender for the use of capital The goal of this book is to provide the mathematical understandings of interest and the time value of money needed to succeed on the actuarial examination covering interest theory Key Features Helps prepare students for the SOA Financial Mathematics Exam Provides mathematical understanding of interest and the time value of money needed to succeed in the actuarial examination covering interest theory Contains many worked examples exercises and solutions for practice Provides training in the use of calculators for solving problems A complete solutions manual is available to faculty adopters online

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