CHEMISTRY COMES ALIVE

Matter - anything with mass and occupies space

- weight pull of gravity on mass
- solid with definite shape and volume

liquid - with definite volume but changeable shape gas - with changeable volume and shape

Energy - capacity to do work or put matter into motion

- 2 types kinetic energy energy in action potential energy – stored (inactive) energy
- energy can be transferred from potential to kinetic

Forms of energy used by the body:

- Chemical energy stored in the chemical bonds
 - ATP (energy of the body)
- 2. Electrical energy result from movement of charged particles
- Mechanical energy energy that directly cause the movement of matter
- Radiant or electromagnetic energy energy that travel in waves.
 - light, infrared waves, radiowaves, untraviolet waves, X-rays

Energy - is neither created nor destroyed

- can be converted from one form to another form,
- some energy is unused in the conversion (lighting a bulb)

Composition of Matters: All matters are made up of elements

Elements: - unique substances that cannot be broken into simpler form by ordinary chemical means

- 2 properties of elements physical anything we can sense or measure
 - chemical how atoms interact (bond) with other atoms

Atoms -unique building blocks of elements

- -give each element its chemical and physical properties
- -smallest particles of the element that retain the properties of the element
- atomic symbols designated for each element.
 - usually represented by the first letter or the first 2 letter of the name of the element in English, Latin mainly and rarely in other languages
 - 118 elements 92 are naturally occurring
 - 26 are produced from natural element by particle accelerator or nuclear reactor.

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