Organic Chemistry II Study Guide for Exam I Chapters 10, 11, and 14

- Classification of Alcohols
 - Alcohols are classified according to how many carbons are attached to the carbon containing the -OH group.
 - Methyl, primary, secondary, and tertiary
- Nomenclature of Alcohols
 - Naming an alcohol
 - Start numbering the carbon chain to reach the -OH group first
 - Include the location (carbon number) of the -OH group



- 2-methylhexan-2-ol
- Naming an alcohol as a substituent
 - Do this when alcohol is NOT the highest priority group (carboxylic acid, etc.)
 - Start numbering the carbon chain at the higher priority group
 - Use the prefix hydroxy- and include the location (carbon number) of the -OH group

- 4-hydroxybutanoic acid
- Naming cyclic alcohols
 - If the -OH is the highest priority group, do NOT include the location (carbon number)
 - · It is implied so it is unnecessary



- 1,4,4-trimethylcyclohexanol
- Naming diols
 - When 2 -OH groups are present and is the highest priority group, us the suffix -diol
 - Include the location (carbon number) of BOTH -OH groups



Propane-1,3-diol

Study Guide For Organic Chemistry

Marc Loudon

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<u>Essential Organic Chemistry Study Guide & Solution Manual, Books a la Carte Edition</u> Paula Yurkanis

Bruice, 2015-05-17

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