

Name \_\_\_\_\_ Date \_\_\_\_\_ Class \_\_\_\_\_

## CHAPTER 12

## STUDY GUIDE FOR CONTENT MASTERY

## Section 12.3 Limiting Reactants

In your textbook, read about why reactions stop and how to determine the limiting reactant.

Study the diagram showing a chemical reaction and the chemical equation that represents the reaction. Then complete the table. Show your calculations for questions 25–27 in the space below the table.



The molar masses of  $\text{O}_2$ ,  $\text{NO}$ , and  $\text{NO}_2$  are 32.00 g/mol, 30.01 g/mol, and 46.01 g/mol, respectively.

| Amount of $\text{O}_2$ | Amount of $\text{NO}$ | Amount of $\text{NO}_2$ | Limiting Reactant | Amount and State of Excess Reactant |
|------------------------|-----------------------|-------------------------|-------------------|-------------------------------------|
| 1 molecule             | 2 molecules           | 2 molecules             | none              | none                                |
| 4 molecules            | 4 molecules           | 4 molecules             | NO                | 2 molecules $\text{O}_2$            |
| 2 molecules            | 8 molecules           | 4 molecules             | 2. $\text{O}_2$   | 3. 4 molecules NO                   |
| 1.00 mol               | 2.00 mol              | 4. 2.00 mol             | 5. none           | 6. none                             |
| 4.00 mol               | 4.00 mol              | 7. 4.00 mol             | 8. NO             | 9. 2.00 mol $\text{O}_2$            |
| 1.00 mol               | 7.00 mol              | 10. 7.00 mol            | 11. NO            | 12. 1.50 mol $\text{O}_2$           |
| 1.00 mol               | 4.00 mol              | 13. 2.00 mol            | 14. $\text{O}_2$  | 15. 2.00 mol NO                     |
| 0.500 mol              | 0.200 mol             | 16. 0.200 mol           | 17. NO            | 18. 0.400 mol $\text{O}_2$          |
| 32.00 g                | 60.02 g               | 19. 62.82 g             | 20. none          | 21. none                            |
| 16.00 g                | 60.02 g               | 22. 46.01 g             | 23. $\text{O}_2$  | 24. 58.12 g NO                      |
| 16.00 g                | 20.00 g               | 25. 28.76 g             | 26. $\text{O}_2$  | 27. 1.24 g NO                       |

balanced equation mole ratio = 2 mol  $\text{NO}$ /1 mol  $\text{O}_2$

$16.00 \text{ g } \text{O}_2 \times 1 \text{ mol } \text{O}_2 / 32.00 \text{ g } \text{O}_2 = 0.5125 \text{ mol } \text{O}_2$

$20.00 \text{ g } \text{NO} \times 1 \text{ mol } \text{NO} / 30.01 \text{ g } \text{NO} = 0.6664 \text{ mol } \text{NO}$

actual mole ratio =  $0.6664 \text{ mol } \text{NO} / 0.5125 \text{ mol } \text{O}_2 = 2.132 \text{ mol } \text{NO} / 1.000 \text{ mol } \text{O}_2$

Because the actual mole ratio of  $\text{NO}:\text{O}_2$  is larger than the balanced equation mole ratio of  $\text{NO}:\text{O}_2$ , there is an excess of  $\text{NO}$ ;  $\text{O}_2$  is the limiting reactant.

Mass of  $\text{NO}$  used =  $0.5125 \text{ mol } \text{O}_2 \times 2 \text{ mol } \text{NO} / 1 \text{ mol } \text{O}_2 = 0.4250 \text{ mol } \text{NO}$

$0.4250 \text{ mol } \text{NO} \times 30.01 \text{ g } \text{NO} / 1 \text{ mol } \text{NO} = 12.76 \text{ g } \text{NO}$

Mass of  $\text{NO}_2$  produced =  $0.4250 \text{ mol } \text{NO}_2 \times 46.01 \text{ g } \text{NO}_2 / 1 \text{ mol } \text{NO}_2 = 19.56 \text{ g } \text{NO}_2$

Excess  $\text{NO} = 20.00 \text{ g } \text{NO} - 12.76 \text{ g } \text{NO} = 7.24 \text{ g } \text{NO}$

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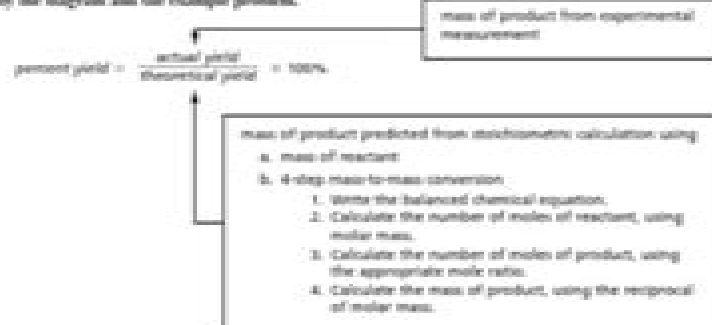
## CHAPTER 12

## STUDY GUIDE FOR CONTENT MASTERY

## Section 12.4 Percent Yield

In your textbook, read about the yield of products.

Study the diagram and the example problem.



**Example Problem:** The following chemical equation represents the production of gallium oxide, a substance used in the manufacturing of some semiconductor devices.



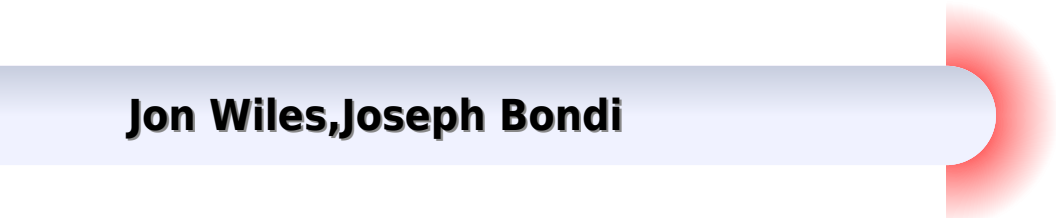
In one experiment, the reaction yielded 7.42 g of the oxide from a 7.00-g sample of gallium. Determine the percent yield of this reaction. The molar masses of Ga and  $\text{Ga}_2\text{O}_3$  are 69.72 g/mol and 187.44 g/mol, respectively.

Use the information in the diagram and example problem to evaluate each value or expression below. If the value or expression is correct, write correct. If it is incorrect, write the correct value or expression.

- actual yield: unknown 7.42 g  $\text{Ga}_2\text{O}_3$
- mass of reactant: 7.00 g Ga correct
- number of moles of reactant:  $7.00 \text{ g } \text{Ga} \times \frac{69.72 \text{ g } \text{Ga}}{1 \text{ mol } \text{Ga}} = 7.00 \text{ g } \text{Ga} \times 1 \text{ mol } \text{Ga} / 69.72 \text{ g } \text{Ga}$
- number of moles of product:  $0.100 \text{ mol } \text{Ga} \times \frac{2 \text{ mol } \text{Ga}_2\text{O}_3}{1 \text{ mol } \text{Ga}} = 0.100 \text{ mol } \text{Ga}_2\text{O}_3 / 4 \text{ mol } \text{Ga}$
- theoretical yield:  $0.100 \text{ mol } \text{Ga}_2\text{O}_3 \times \frac{187.44 \text{ g } \text{Ga}_2\text{O}_3}{1 \text{ mol } \text{Ga}_2\text{O}_3} = 18.74 \text{ g } \text{Ga}_2\text{O}_3$  correct
- percent yield:  $\frac{7.42 \text{ g } \text{Ga}_2\text{O}_3}{18.74 \text{ g } \text{Ga}_2\text{O}_3} \times 100 = 7.42 \text{ g } \text{Ga}_2\text{O}_3 / 18.74 \text{ g } \text{Ga}_2\text{O}_3 \times 100$

# Study Guide For Content Mastery Percent Yield

**Jon Wiles, Joseph Bondi**



## **Study Guide For Content Mastery Percent Yield:**

**The Nurse Educators Guide to Assessing Learning Outcomes** McDonald, 2017-07-28 The Nurse Educators Guide to Assessing Learning Outcomes Fourth Edition is a widely used resource for both faculty and nursing education students that covers the assessment of critical thinking the development of learning objectives and the creation of tests including detailed tips for writing many kinds of individual test items The book also covers the analysis of test reliability Important Notice The digital edition of this book is missing some of the images or content found in the physical edition Guide to the Setting of Appropriate Cutting Scores for Written Tests Lawrence S. Buck, 1977 **Certified Nurse Educator (CNE) Review Manual** Maryann Godshall, Ruth A. Wittmann-Price, 2009-06-15 Nurse Educators can no longer be just experts in their specific fields they need to be able to teach a variety of students in a variety of settings This book encompasses all topics necessary to assist nurses to become educators I am not aware any other books that address this area so thoroughly Score 98 5 stars Doody's I truly believe that your book made it possible for me to successfully complete the CNE exam It truly is the only book you need to pass the CNE exam I'm living proof Thanks so much I learned a lot from your book and intend to keep it on my desk for future reference Sue Carroll MSN RN Sentara College of Health Sciences This is the only Certified Nurse Educator CNE examination prep book that addresses each of the key topics outlined in the actual NLN test blueprint The review manual serves as an indispensable resource for novice and expert nurse educators seeking to master the content needed for CNE Certification Each chapter contains only the most important information needed for the test breaking each topic down into its parts to facilitate quick access to core content No other review manual offers comprehensive coverage of everything nurse educators need to know for the CNE exam including guidelines for classroom management practical teaching strategies and much more Why you need this book Chapter content presents nurse educator core competencies as outlined in the NLN test blueprint Features practice questions in each chapter and an answer key in the back of the book with full explanations of correct answers Includes comprehensive practice test at the end of the book also with explanations of correct answers Contains essential information on practical teaching strategies developing lesson plans culturally diverse students student evaluation and much more This guide contains the essential knowledge that all nurse educators need to pass the test and excel in teaching thus serving as an invaluable resource for systematic review of content before taking the NLN CNE examination State Manual of the Courses of Study for the High Schools of Oregon Oregon. Office of Superintendent of Public Instruction, 1927 **Testing, Teaching, and Learning** National Research Council, Division of Behavioral and Social Sciences and Education, Board on Testing and Assessment, Committee on Title I Testing and Assessment, 1999-10-06 State education departments and school districts face an important challenge in implementing a new law that requires disadvantaged students to be held to the same standards as other students The new requirements come from provisions of the 1994 reauthorization of Title I the largest federal effort in precollegiate education which provides aid to level the field for

disadvantaged students Testing Teaching and Learning is written to help states and school districts comply with the new law offering guidance for designing and implementing assessment and accountability systems This book examines standards based education reform and reviews the research on student assessment focusing on the needs of disadvantaged students covered by Title I With examples of states and districts that have track records in new systems the committee develops a practical decision framework for education officials The book explores how best to design assessment and accountability systems that support high levels of student learning and to work toward continuous improvement Testing Teaching and Learning will be an important tool for all involved in educating disadvantaged students state and local administrators and classroom teachers

**Adapting Unstoppable Learning** Yazmin Pineda Zapata, Rebecca Brooks, 2017-05-30 Foreword by Douglas Fisher and Nancy Frey Modify your curriculum using streamlined assignments targeted assessments and student engagement strategies for different learning styles This practical guide expands upon the Unstoppable Learning model to explore accessible learning for all students Through specific curriculum and environmental accommodations and modifications as well as personal and technology supports K 12 teachers will discover how to provide differentiated instruction to students with varying needs from physical disabilities to twice exceptionality Forms tools and diagrams designed to aid instructional planning are also included How will this book help you Learn how to adjust curriculum in ways that maintain appropriate levels of rigor for different learning styles Consider real school vignettes and examples that illustrate successful Unstoppable Learning adaptations for inclusive classrooms Study the four guiding principles of systems thinking relationships communications responsiveness and sustainability Consider the importance of collaboration and communication in learning adaptations Study helpful and informative tools and diagrams intended to aid in planning instruction for different learning styles

**Learning Patterns in Higher Education** David Gijbels, Vincent Donche, John T. E. Richardson, Jan D. Vermunt, 2013-09-11 Learning Patterns in Higher Education brings together a cutting edge international team of contributors to critically review our current understanding of how students and adults learn how differences and changes in the way students learn can be measured in a valid and reliable way and how the quality of student learning may be enhanced There is substantial evidence that students in higher education have a characteristic way of learning sometimes called their learning orientation Biggs 1988 learning style Evans et al 2010 or learning pattern Vermunt and Vermetten 2004 However recent research in the field of student learning has resulted in multi faceted and sometimes contradictory results which may reflect conceptual differences and differences in measurement of student learning in each of the studies This book deals with the need for further clarification of how students learn in higher education in the 21st century and to what extent the measurements often used in learning pattern studies are still up to date or can be advanced with present methodological and statistical insights to capture the most important differences and changes in student learning The contributions in the book are organized in two parts a first conceptual and psychological part in which the

dimensions of student learning in the 21st century are discussed and a second empirical part in which questions related to how students learning can be measured and how it develops are considered Areas covered include Cultural influences on learning patterns Predicting learning outcomes Student centred learning environments and self directed learning Mathematics learning This indispensable book covers multiple conceptual perspectives on how learning patterns can be described and effects and developments can be measured and will not only be helpful for learning researchers as such but also for educational researchers from the broad domain of educational psychology motivation psychology and instructional sciences who are interested in student motivation self regulated learning effectiveness of innovative learning environments as well as assessment and evaluation of student characteristics and learning process variables **A Road Map to PLC**

**Success** Sean McWherter, 2017-03-16 Educators often have trouble properly implementing Professional Learning Communities PLCs because they simply don't know how the process is supposed to work By cutting through the fluff and generalities this book provides a clear road map that takes school leaders step by step through the entire PLC process Each chapter addresses a foundational component or protocol necessary for building successful team based learning communities using real life examples to help teachers and leaders understand how to integrate this process and avoid common pitfalls that inhibit implementation Whether you're just starting the PLC process or you're looking to get more out of your PLCs this book will lead you to continued student and teacher growth regardless of current achievement levels socioeconomic status or impending curricular changes **Handbook on Measurement, Assessment, and Evaluation in Higher Education** Charles

Secolsky, D. Brian Denison, 2017-07-31 In this valuable resource well known scholars present a detailed understanding of contemporary theories and practices in the fields of measurement assessment and evaluation with guidance on how to apply these ideas for the benefit of students and institutions Bringing together terminology analytical perspectives and methodological advances this second edition facilitates informed decision making while connecting the latest thinking in these methodological areas with actual practice in higher education This research handbook provides higher education administrators student affairs personnel institutional researchers and faculty with an integrated volume of theory method and application

**Teaching at Its Best** Linda B. Nilson, 2010-04-20 Teaching at Its Best This third edition of the best selling handbook offers faculty at all levels an essential toolbox of hundreds of practical teaching techniques formats classroom activities and exercises all of which can be implemented immediately This thoroughly revised edition includes the newest portrait of the Millennial student current research from cognitive psychology a focus on outcomes maps the latest legal options on copyright issues and how to best use new technology including wikis blogs podcasts vodcasts and clickers Entirely new chapters include subjects such as matching teaching methods with learning outcomes inquiry guided learning and using visuals to teach and new sections address Felder and Silverman's Index of Learning Styles SCALE UP classrooms multiple true false test items and much more Praise for the Third Edition of Teaching at Its Best Everyone veterans as well as

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