

Name: _____

Atomic Structure & the Periodic Table Study Guide

Atomic Structure

Particle	Charge	Where are they found?
Proton	Positive	Nucleus
Neutron	Neutral/ no charge	Nucleus
Electron	negative	Electron cloud/energy levels outside nucleus

1. What is the rule for writing chemical symbols like Na or Ca? **Capitalize the first letter; lower case second letter**
2. What are the vertical columns on the periodic table called? **Groups** or **families**
 - a. How many of these are on the periodic table? **18**
 - b. Why are the elements in these columns? **They have the same chemical properties** (& same # of valence electrons)
 - c. Elements in the same group have the same number of **valence** electrons.
3. What are the horizontal rows on the periodic table called? **periods**
 - a. How many of these are on the periodic table? **7**
 - b. Elements in these rows have the same number of **energy** (electron shells) levels.
4. What are valence electrons? **Electrons in the outer energy level**
5. How many electrons are in each energy level?
1st energy level **2** 2nd energy level **8** 3rd energy level **8**
6. The atomic number is equal to the **# of protons** and also **# of electrons**
7. The atomic mass is equal to the **# protons + # of neutrons**
8. How do you find the number of neutrons? **Atomic mass – atomic number**
9. We view the order of the periodic table today according to the **atomic number**

Study Guide For Periodic Table

Pamela Mills, Amina El-Ashmawy



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Periodic Table (Advanced) Speedy Publishing,2014-08-01 An advanced periodic table of elements displays not only the elements but the ions that form each element A pamphlet with such a visual aid would greatly benefit chemistry students Any student taking chemistry will need to learn the elements A pamphlet would be concise and break the information down simply making it easier to understand and remember It allows students to simply focus on the main point rather than taking in information that they may or may not need *Know Your 'O' Level Chemistry - A Study Guide* , [Barron's Science 360: A Complete Study Guide to Chemistry with Online Practice](#) Barron's Educational Series,Mark Kernion,Joseph A.

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depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane CH₄ from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement

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Table of Contents Study Guide For Periodic Table

1. Understanding the eBook Study Guide For Periodic Table
 - The Rise of Digital Reading Study Guide For Periodic Table
 - Advantages of eBooks Over Traditional Books
2. Identifying Study Guide For Periodic Table
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Study Guide For Periodic Table
 - User-Friendly Interface
4. Exploring eBook Recommendations from Study Guide For Periodic Table
 - Personalized Recommendations
 - Study Guide For Periodic Table User Reviews and Ratings
 - Study Guide For Periodic Table and Bestseller Lists
5. Accessing Study Guide For Periodic Table Free and Paid eBooks
 - Study Guide For Periodic Table Public Domain eBooks
 - Study Guide For Periodic Table eBook Subscription Services
 - Study Guide For Periodic Table Budget-Friendly Options

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 - Highlighting and Note-Taking Study Guide For Periodic Table
 - Interactive Elements Study Guide For Periodic Table
8. Staying Engaged with Study Guide For Periodic Table
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Study Guide For Periodic Table
9. Balancing eBooks and Physical Books Study Guide For Periodic Table
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Study Guide For Periodic Table
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Study Guide For Periodic Table
 - Setting Reading Goals Study Guide For Periodic Table
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Study Guide For Periodic Table
 - Fact-Checking eBook Content of Study Guide For Periodic Table
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

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