

# CHAPTER 14 STUDY GUIDE

## Mixtures and Solutions

### Section 14.1 Heterogeneous and Homogeneous Mixtures

In your textbook, read about suspensions and colloids.

For each statement below, write *true* or *false*.

- |                |                                                                                                                                   |
|----------------|-----------------------------------------------------------------------------------------------------------------------------------|
| <u> false </u> | 1. A solution is a mixture containing particles that settle out of the mixture if left undisturbed.                               |
| <u> true </u>  | 2. The most abundant substance in a colloid is the dispersion medium.                                                             |
| <u> false </u> | 3. A colloid can be separated by filtration.                                                                                      |
| <u> true </u>  | 4. A solid emulsion consists of a liquid dispersed in a solid.                                                                    |
| <u> true </u>  | 5. Whipped cream is an example of a foam.                                                                                         |
| <u> false </u> | 6. In an aerosol, the dispersing medium is a liquid.                                                                              |
| <u> true </u>  | 7. Brownian motion results from the collisions of particles of the dispersion medium with the dispersed particles.                |
| <u> true </u>  | 8. Dispersed particles in a colloid do not tend to settle out because they have polar or charged atomic groups on their surfaces. |
| <u> false </u> | 9. Stirring an electrolyte into a colloid stabilizes the colloid.                                                                 |
| <u> true </u>  | 10. Colloids demonstrate the Tyndall effect.                                                                                      |

The table below lists the characteristics of particles in colloids, solutions, and suspensions. Place a check in the column of each mixture whose particles have a particular characteristic.

| Characteristics of Particles             | Colloid                             | Solution                            | Suspension                          |
|------------------------------------------|-------------------------------------|-------------------------------------|-------------------------------------|
| 11. Less than 1 nm in diameter           |                                     | <input checked="" type="checkbox"/> | <input checked="" type="checkbox"/> |
| 12. Between 1 nm and 1000 nm in diameter | <input checked="" type="checkbox"/> | <input checked="" type="checkbox"/> | <input checked="" type="checkbox"/> |
| 13. More than 1000 nm in diameter        | <input checked="" type="checkbox"/> | <input checked="" type="checkbox"/> |                                     |
| 14. Settle out if undisturbed            |                                     |                                     |                                     |
| 15. Pass through standard filter paper   | <input checked="" type="checkbox"/> |                                     |                                     |
| 16. Lower vapor pressure                 |                                     |                                     | <input checked="" type="checkbox"/> |
| 17. Scatter light                        | <input checked="" type="checkbox"/> |                                     |                                     |

# Study Guide Mixtures And Solutions Answer Key

**G Psacharopoulos**



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