

# Unit 10



# STUDY GUIDE FOR CONCEPT MASTERY

## Section 8.1 (continued)

In your notebook, read aloud the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

|       |
|-------|
| _____ |
| _____ |
| _____ |
| _____ |
| _____ |

|                        |       |
|------------------------|-------|
| atomic mass            | _____ |
| atomic number          | _____ |
| electron configuration | _____ |
| chemical name          | _____ |
| chemical symbol        | _____ |

Put each item in Column A, either the letter of the matching item in Column B.

### Column A

- \_\_\_\_\_ 18. A column in the periodic table
- \_\_\_\_\_ 19. A row in the periodic table
- \_\_\_\_\_ 20. Group 1 elements
- \_\_\_\_\_ 21. Elements that are strong oxidizing agents
- \_\_\_\_\_ 22. Group 8 elements

### Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write *yes* if the statement is true, if the statement is false, change the underlined word or phrase to make it true.

- \_\_\_\_\_ 23. There are two representative elements of phosphorus.
- \_\_\_\_\_ 24. Most transition metals of the elements in the periodic table are metalloids.
- \_\_\_\_\_ 25. Group 1A elements except for hydrogen are metals in the solid state.
- \_\_\_\_\_ 26. Group 2A elements are the alkaline earth metals.
- \_\_\_\_\_ 27. Group 7A elements are highly reactive nonmetals known as halogens.
- \_\_\_\_\_ 28. Group 8A elements are inert nonmetallic elements known as noble gases.
- \_\_\_\_\_ 29. Alkaloids have properties of both metals and nonmetals.

# Study Guide For Content Mastery Physical Oceanography

**Michael Hoelscher**



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