

10.1 Periodic Trends



PERIODIC TRENDS: FROM CONCEPT TO PRACTICE

Section 10.1 continued

In your notebook, read aloud the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

_____	1
_____	2
_____	3
_____	4
_____	5
_____	6
_____	7
_____	8
_____	9
_____	10
_____	11
_____	12
_____	13
_____	14
_____	15
_____	16
_____	17
_____	18

Atomic Mass	_____
Atomic Number	_____
Electron Configuration	_____
Chemical Name	_____
Chemical Symbol	_____

Put each item in Column A, either the letter of the matching item in Column B.

Column A

- _____ 19. It increases in the periodic table.
- _____ 20. It decreases in the periodic table.
- _____ 21. Group 1 elements.
- _____ 22. Elements that are shiny and conduct electricity.
- _____ 23. Group 17 elements.

Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write each of the statements in terms of the statement in bold. Change the underlined word or phrase to make it true.

- _____ 24. There are two representative elements of elements.
- _____ 25. Most elements are metals in the periodic table.
- _____ 26. Group 1A elements except for hydrogen are metals in the periodic table.
- _____ 27. Group 2A elements are the alkaline earth metals.
- _____ 28. Group 7A elements are highly reactive nonmetals known as halogens.
- _____ 29. Group 8A elements are inert nonreactive elements known as noble gases.
- _____ 30. Elements from groups of both metals and nonmetals are metalloids.

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AW Chickering



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