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Student Exploration: Ionic Bonds

Directions: Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

Vocabulary: chemical family, electron affinity, ion, ionic bond, metal, nonmetal, octet rule, shell, valence electron

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. Nate and Clara are drawing pictures with markers. There are 8 markers in a set. Nate has 9 markers and Clara has 7. What can Nate and Clara do so that each of them has a full set?

Nate can give Clara 1 marker, so they each have a full set.

2. Maggie is sitting at a table with Fred and Florence. Maggie has 10 markers, but Fred and Florence each have only 7 markers. How can they share markers so each has 8?

Maggie has to give away 2 markers to Fred and Florence so each has 8.

Gizmo Warm-up

Just like students sharing markers, atoms sometimes share or swap electrons. By doing this, atoms form bonds. The *Ionic Bonds* Gizmo allows you to explore how **ionic bonds** form.

To begin, check that **Sodium (Na)** and **Chlorine (Cl)** are selected from the menus at right. Click **Play** (▶) to see electrons orbiting the nucleus of each atom. (Note: These atom models are simplified and not meant to be realistic.)



1. Each atom consists of a central nucleus and several **shells** that contain electrons. The outermost electrons are called **valence electrons**. (Inner electrons are not shown.)

How many valence electrons does each atom have? Sodium: 1 Chlorine: 7

2. Click **Pause** (⏸). Elements can be classified as **metals** and **nonmetals**. Metals do not hold on to their valence electrons very tightly, while nonmetals hold their electrons tightly. **Electron affinity** is a measure of how tightly the valence electrons are held.

- A. Try pulling an electron away from each atom. Based on this experiment, which atom is a metal?

Sodium

Which is a nonmetal?

Chlorine

- B. Try moving an electron from the metal to the nonmetal. What happens?

Student Exploration Ionic Bonds Gizmo Answer Key

KJ Lindholm-Leary

A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the author's name.

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Table of Contents Student Exploration Ionic Bonds Gizmo Answer Key

1. Understanding the eBook Student Exploration Ionic Bonds Gizmo Answer Key
 - The Rise of Digital Reading Student Exploration Ionic Bonds Gizmo Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Student Exploration Ionic Bonds Gizmo Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Student Exploration Ionic Bonds Gizmo Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Student Exploration Ionic Bonds Gizmo Answer Key
 - Personalized Recommendations
 - Student Exploration Ionic Bonds Gizmo Answer Key User Reviews and Ratings
 - Student Exploration Ionic Bonds Gizmo Answer Key and Bestseller Lists

5. Accessing Student Exploration Ionic Bonds Gizmo Answer Key Free and Paid eBooks
 - Student Exploration Ionic Bonds Gizmo Answer Key Public Domain eBooks
 - Student Exploration Ionic Bonds Gizmo Answer Key eBook Subscription Services
 - Student Exploration Ionic Bonds Gizmo Answer Key Budget-Friendly Options
6. Navigating Student Exploration Ionic Bonds Gizmo Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Student Exploration Ionic Bonds Gizmo Answer Key Compatibility with Devices
 - Student Exploration Ionic Bonds Gizmo Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Student Exploration Ionic Bonds Gizmo Answer Key
 - Highlighting and Note-Taking Student Exploration Ionic Bonds Gizmo Answer Key
 - Interactive Elements Student Exploration Ionic Bonds Gizmo Answer Key
8. Staying Engaged with Student Exploration Ionic Bonds Gizmo Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Student Exploration Ionic Bonds Gizmo Answer Key
9. Balancing eBooks and Physical Books Student Exploration Ionic Bonds Gizmo Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Student Exploration Ionic Bonds Gizmo Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Student Exploration Ionic Bonds Gizmo Answer Key
 - Setting Reading Goals Student Exploration Ionic Bonds Gizmo Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Student Exploration Ionic Bonds Gizmo Answer Key
 - Fact-Checking eBook Content of Student Exploration Ionic Bonds Gizmo Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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