

Name: Daniela Martinez

Date: _____

Student Exploration: Stoichiometry

Vocabulary: Avogadro's number, balanced equation, cancel, coefficient, conversion factor, dimensional analysis, molar mass, mole, molecular mass, stoichiometry

Activity A: Moles

Get the Gizmo ready:

- Check that the equation is still:
$$\text{Fe}_2\text{O}_3 + 3\text{CO} \rightarrow 2\text{Fe} + 3\text{CO}_2$$
- If not, click **New equation** until it reappears.



Question: How do scientists find the molecular mass and molar mass of a substance?

- Calculate:** The **molecular mass** of a molecule is the sum of the masses of each atom in the molecule. The unit of molecular mass is the universal mass unit (u).

Iron's atomic mass is 55.85 u, carbon's mass is 12.01 u, and oxygen's mass is 16.00 u.

- Calculate the molecular mass of carbon monoxide (CO) by adding the atomic mass of carbon and the atomic mass of oxygen: 28.01g
- Calculate the molecular mass of carbon dioxide (CO₂): 44.01g
- Calculate the molecular mass of iron (III) oxide (Fe₂O₃): 159.7g

- Infer:** A mole of a substance has a mass in grams that is equal to the molecular mass. For example, a carbon atom has a mass of 12.01 u. A mole of carbon has a mass of 12.01 g. Based on their molecular masses, list the **molar mass** of each substance:

Fe₂O₃ 159.7g CO 28.01g Fe 55.85g CO₂ 44.01g

Check your answers on the Gizmo by inspecting the middle row of tiles on the right side of the Gizmo. These tiles show the units "1 mol" on top and "g" below.

- Practice:** Hydrogen has an atomic mass of 1.01 u. What is the molar mass of these substances?

H₂O 18.02g CH₄ 16.05g H₂CO₃ 62.03g C₆H₁₂O₆ 180.18g

Student Exploration Stoichiometry Explorelearning

Sydney Young



Student Exploration Stoichiometry Explorelearning:

Handbook of Educational Data Mining Cristobal Romero, Sebastian Ventura, Mykola Pechenizkiy, Ryan S.J.d. Baker, 2010-10-25 This handbook provides a thorough overview of the current state of knowledge in this area The first part of the book includes nine surveys and tutorials on the principal data mining techniques that have been applied in education The second part presents a set of 25 case studies that give a rich overview of the problems that EDM has addressed With contributions by well known researchers from a variety of fields the book reflects the multidisciplinary nature of the EDM community It helps education experts understand what types of questions EDM can address and helps data miners understand what types of questions are important to educational design and educational decision making **Stoichiometry** Sydney Young, 1908 *You Can Do Chemistry* Sunil Tanna, 2018-11-19 A comprehensive guide to performing mole and stoichiometric calculations with numerous examples as well as questions and answers Covers calculations relating to solids solutions gases and electrolysis plus as limiting and excess reactants chemical yields atom economy and much more Fully up to date with the last international standards including the revised definition of mole which was agreed on November 16th 2018

This book delves into Student Exploration Stoichiometry Explorelearning. Student Exploration Stoichiometry Explorelearning is an essential topic that needs to be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Student Exploration Stoichiometry Explorelearning, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:
 - Chapter 1: Introduction to Student Exploration Stoichiometry Explorelearning
 - Chapter 2: Essential Elements of Student Exploration Stoichiometry Explorelearning
 - Chapter 3: Student Exploration Stoichiometry Explorelearning in Everyday Life
 - Chapter 4: Student Exploration Stoichiometry Explorelearning in Specific Contexts
 - Chapter 5: Conclusion
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 3. In chapter 2, this book will delve into the foundational concepts of Student Exploration Stoichiometry Explorelearning. The second chapter will elucidate the essential principles that must be understood to grasp Student Exploration Stoichiometry Explorelearning in its entirety.
 4. In chapter 3, the author will examine the practical applications of Student Exploration Stoichiometry Explorelearning in daily life. This chapter will showcase real-world examples of how Student Exploration Stoichiometry Explorelearning can be effectively utilized in everyday scenarios.
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 6. In chapter 5, this book will draw a conclusion about Student Exploration Stoichiometry Explorelearning. This chapter will summarize the key points that have been discussed throughout the book.
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