

CHEMISTRY (SCIENCE): STOICHIOMETRIC CALCULATIONS (MOLE)

HARVARD UNIVERSITY

Study guide: Definitions, Explanation, Examples and Notes

Calculations from

equations

Mole to

mole calculations

Examples

1. How many moles of chlorine are required to react 2.5 moles of calcium to produce calcium chloride?

Solution



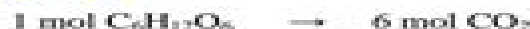
$$x = \frac{2.5 \text{ mol Ca} \times 1 \text{ mol Cl}_2}{1 \text{ mol Ca}}$$

$$x = 2.5 \text{ mol}$$

2. How many moles of carbon dioxide (CO₂) will be produced by complete combustion of 2 moles of glucose (C₆H₁₂O₆) according to the equation?



Solution



$$x = \frac{2 \text{ mol C}_6\text{H}_{12}\text{O}_6 \times 6 \text{ mol CO}_2}{1 \text{ mol C}_6\text{H}_{12}\text{O}_6}$$

$$x = 12 \text{ mol CO}_2$$

Mole to mass

calculations

Examples

Study Guide Chemistry Stoichiometric Calculations

Rosina Ehmann



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