Name: Date: 22 Jan 20

Student Exploration: Ionic Bonds

Vocabulary: chemical family, electron affinity, ion, ionic bond, metal, nonmetal, octet rule, shell, valence electron, sheavy metal s

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

Nate and Clara are drawing pictures with markers. There are 8 markers in a set. Nate has 9
markers and Clara has 7. What can Nate and Clara do so that each of them has a full set?

Nate gives one to Clara

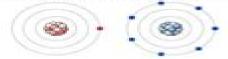
Maggie is sitting at a table with Fred and Florence. Maggie has 10 markers, but Fred and Florence each have only 7 markers. How can they share markers so each has 8?

Maggies gives Fred one marker and Maggie gives Florence one marker.

Gizmo Warm-up

Just like students sharing markers, atoms sometimes share or swap electrons. By doing this, atoms form bonds. The lonic Bonds Gizmo™ allows you to explore how **ionic bonds** form.

To begin, check that **Sodium** (Na) and **Chlorine** (Cl) are selected from the menus at right. Click **Play** () to see electrons orbiting the nucleus of each atom. (Note: These atom models are simplified and not meant to be realistic.)



 Each atom consists of a central nucleus and several shells that contain electrons. The outermost electrons are called valence electrons. (Inner electrons are not shown.)

How many valence electrons does each atom have? Sodium:_

odium:___1___ Chlorine:___7

- Click Pause (). Elements can be classified as metals and nonmetals. Metals do not hold on to their valence electrons very tightly, while nonmetals hold their electrons tightly. Electron affinity is a measure of how tightly the valence electrons are held.
 - Try pulling an electron away from each atom. Based on this

experiment, which atom is a metal? SodiumWhich is a nonmetal?

Chlorine

B. Try moving an electron from the metal to the nonmetal. What happens?

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Kevin W. Plaxco, Michael Gross

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Table of Contents Student Exploration Ionic Bonds

- 1. Understanding the eBook Student Exploration Ionic Bonds
 - The Rise of Digital Reading Student Exploration Ionic Bonds
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Student Exploration Ionic Bonds
 - Exploring Different Genres
 - o Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - $\circ \ \ Popular \ eBook \ Platforms$
 - Features to Look for in an Student Exploration Ionic Bonds
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Student Exploration Ionic Bonds
 - Personalized Recommendations

- Student Exploration Ionic Bonds User Reviews and Ratings
- Student Exploration Ionic Bonds and Bestseller Lists
- 5. Accessing Student Exploration Ionic Bonds Free and Paid eBooks
 - Student Exploration Ionic Bonds Public Domain eBooks
 - Student Exploration Ionic Bonds eBook Subscription Services
 - Student Exploration Ionic Bonds Budget-Friendly Options
- 6. Navigating Student Exploration Ionic Bonds eBook Formats
 - ∘ ePub, PDF, MOBI, and More
 - Student Exploration Ionic Bonds Compatibility with Devices
 - Student Exploration Ionic Bonds Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Student Exploration Ionic Bonds
 - Highlighting and Note-Taking Student Exploration Ionic Bonds
 - Interactive Elements Student Exploration Ionic Bonds
- 8. Staying Engaged with Student Exploration Ionic Bonds
 - o Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Student Exploration Ionic Bonds
- 9. Balancing eBooks and Physical Books Student Exploration Ionic Bonds
 - ∘ Benefits of a Digital Library
 - Creating a Diverse Reading Collection Student Exploration Ionic Bonds
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Student Exploration Ionic Bonds
 - Setting Reading Goals Student Exploration Ionic Bonds
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Student Exploration Ionic Bonds
 - Fact-Checking eBook Content of Student Exploration Ionic Bonds

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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