

Name: _____ Date: 22 Jan 20

Student Exploration: Ionic Bonds

Vocabulary: chemical family, electron affinity, ion, ionic bond, metal, nonmetal, octet rule, shell, valence electron, Δ heavy metal Δ

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. Nate and Clara are drawing pictures with markers. There are 8 markers in a set. Nate has 9 markers and Clara has 7. What can Nate and Clara do so that each of them has a full set?

Nate gives one to Clara

2. Maggie is sitting at a table with Fred and Florence. Maggie has 10 markers, but Fred and Florence each have only 7 markers. How can they share markers so each has 8?

Maggies gives Fred one marker and Maggie gives Florence one marker.

Gizmo Warm-up

Just like students sharing markers, atoms sometimes share or swap electrons. By doing this, atoms form bonds. The Ionic Bonds Gizmo™ allows you to explore how **ionic bonds** form.

To begin, check that **Sodium (Na)** and **Chlorine (Cl)** are selected from the menus at right. Click **Play** () to see electrons orbiting the nucleus of each atom. (Note: These atom models are simplified and not meant to be realistic.)



1. Each atom consists of a central **nucleus** and several **shells** that contain electrons. The outermost electrons are called **valence electrons**. (Inner electrons are not shown.)

How many valence electrons does each atom have? Sodium: 1 Chlorine: 7

2. Click **Pause** (). Elements can be classified as **metals** and **nonmetals**. Metals do not hold on to their valence electrons very tightly, while nonmetals hold their electrons tightly. **Electron affinity** is a measure of how tightly the valence electrons are held.

- A. Try pulling an electron away from each atom. Based on this experiment, which atom is a metal? Sodium Which is a nonmetal? Chlorine

- B. Try moving an electron from the metal to the nonmetal. What happens?

Student Exploration Ionic Bonds

Kevin W. Plaxco, Michael Gross



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Student-oriented Program National Science Foundation (U.S.). Office of Experimental Projects and Programs, 1974

Astrobiology Kevin W. Plaxco, Michael Gross, 2011-09-01 Informed by new planetary discoveries and the findings from recent robotic missions to Mars, Jupiter, and Saturn, scientists are rapidly replacing centuries of speculation about potential extraterrestrial habitats with real knowledge about the possibility of life outside our own biosphere, if it exists, and where. This second edition of Kevin W. Plaxco and Michael Gross's widely acclaimed text incorporates the latest research in astrobiology to bring readers the most comprehensive up-to-date and engaging introduction to the field available. Plaxco and Gross expand their examination of the origin of chemical elements, the developments that made the Universe habitable, and how life continues to be sustained. They discuss in great detail the formation of the first galaxies and stars, the diverse chemistry of the primordial planet, the origins of metabolism, the evolution of complex organisms, and the feedback regulation of Earth's climate. They also explore life in extreme habitats, potential extraterrestrial habitats, and the current status of the search for extraterrestrial life. Weaving together the relevant threads of astronomy, geology, chemistry, biophysics, and microbiology, this broadly accessible introductory text captures the excitement, controversy, and progress of the dynamic young field of astrobiology. New to this edition is a glossary of terms and an epilogue recapping the key unanswered questions making *Astrobiology* an ideal primer for students and indeed for anyone curious about life and the Universe.

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Prentice Hall Science Explorer Michael J. Padilla, 2002 **The Science Teacher**, 2004

Teaching Naked José Antonio Bowen, 2012-07-03 You've heard about flipping your classroom; now find out how to do it. Introducing a new way to think about higher education learning and technology that prioritizes the benefits of the human dimension, Jos Bowen recognizes that technology is profoundly changing education and that if students are going to continue to pay enormous sums for campus classes, colleges will need to provide more than what can be found

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