

Chemical Equilibrium

PRINCIPAL FACTS AND IDEAS

- The principles of thermodynamics determine the state of chemical equilibrium for any reaction.
- The equilibrium constant expression of elementary chemistry is equal to a constant at constant temperature when it is expressed in terms of activities.
- The principle of Le Châtelier can predict how a chemical system at equilibrium responds to changes in temperature, pressure, or amounts of substances.
- The coupling of biochemical reactions can be understood through thermodynamics and the use of postulated mechanisms.

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